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Paula D. Hopper



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that many chemistry students can sympathize with but gives the wrong expression for the angular

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web 1 draw five protons in the nucleus of the atom label them with their charge 2 draw six neutrons in the nucleus of the atom 3 draw two electrons in the first energy level and label them with their charge 4 draw three electrons in the second energy level and label them with their charge 5 what element is represented by the diagram

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web may 18 2021 figure pageindex 1 the structure of the atom atoms have protons and neutrons in the center making the nucleus while the electrons orbit the nucleus the modern atomic theory states that atoms of one element are the same while atoms of different elements are different

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web atomic structure practice test a electrons and protons b electrons and neutrons c neutrons and positrons d neutrons and protons the major portion of an atom's mass consists of a alpha particle b beta particle c neutron d electron

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web a carbon 12 12 c b this atom contains six protons and six neutrons there are six electrons in a neutral 12 c atom the net charge of such a neutral atom is zero and the mass number is 12 c the preceding answers are correct d the atom will be stable since c 12 is a stable isotope of carbon e the preceding answer is correct

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web the atomic number tells you the number of of an n one atom of an element it also tells you the number of in a neutral atom of that element the atomic number gives the identity of an element as well as its location on the periodic table no two

different elements will have the give the symbol and number of protons in one atom of atomic number

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web mar 6 2017 1 a beryllium atom has 4 protons 5 neutrons and 4 electrons what is the mass number of this atom 4 5 8 9 13 the mass number is the number of protons and neutrons in the nucleus for this question you add the 4 protons and 5 neutrons to get an answer of 9 2 the lowest principal quantum number for an electron is 0 1 2 3

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web a both the electron geometry and the molecular structure are octahedral b both the electron geometry and the molecular structure are trigonal bipyramid c both the electron geometry and the molecular structure are linear d both the electron geometry and the molecular structure are trigonal planar

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web a an oxygen atom with 8 protons and 8 neutrons b a potassium atom with 19 protons and 20 neutrons c a lithium atom with 3 protons and 4 neutron give complete symbols of each atom including the atomic number and the mass number

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web exams prep master atomic structure incorporates three elements that wholly make an atom and they are named electron proton and neutron in the structure of an atom the nucleus sets in the center and it contains the proton whereas electrons lie on the periphery of the atom in an orbital way the first atomic model was given by j j thomson

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