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web may 15 2020 toluene 1 oxidation with kmno4 mechanism 2 benzoic acid $\text{C}_6\text{H}_5\text{COOH}$ SOCl_2 3 benzoyl chloride $\text{C}_6\text{H}_5\text{COCl}$ LiAlH_4 $\text{CH}_3\text{CH}_2\text{CH}_2\text{CHO}$ $\text{C}_6\text{H}_5\text{CHO}$ $\text{CH}_2\text{OHCH}_2\text{OH}$ h step 1 toluene oxidation with kmno4 kmno4 is a strong oxidising agent which oxidise toluene to benzaldehyde in 1st step

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various parameters have been optimized the oxidation of toluene by aq KMnO_4 gives benzoic acid scheme 1 KMnO_4 in turn is reduced to MnO_2 the reaction did not give any other byproduct