

TITRATION - I

STANDARDISATION OF EDTA

Standard hard water Vs EDTA

Indicator: EBT

S.No.	Volume of standard hard water (ml)	Burette Reading		Volume of EDTA (ml)	Concordant value (ml)
		Initial (ml)	Final (ml)		
1	20	0	10.1	10.1	10.1
2	20	0	10.1	10.1	

Calculation:

Volume of Standard hard water	V_1	=	20ml
Strength of Standard hard water	N_1	=	0.0125 N
Volume of EDTA	V_2	=	10.1 ml
Strength of EDTA	N_2	=?

According to the law of volumetric analysis

$$\begin{aligned}
 V_1 N_1 &= V_2 N_2 \\
 N_2 &= \frac{V_1 \times N_1}{V_2} \\
 &= \frac{20 \times 0.0125}{10.1} \\
 &= \frac{0.25}{10.1} \\
 \text{Strength of EDTA} &= 0.0125 \text{ N}
 \end{aligned}$$

(ii) DETERMINATION OF HARDNESS

Pipette out 20ml of a solution and add 4-5 drops of indicator with EDTA. Repeat the titration for concordance.

(iii) DETERMINATION OF HARDNESS

Take 100ml of hard water. Cool, filter it into a 250 ml flask. Add 20 ml of this solution to EDTA. consumed correct value be V_2 ml. Find from total hardness.

Pkgacp Che 222 Lab Manual

SB Merriam



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