

Chem 125A Pre-Lab Quiz 1 Section _____ Name _____

- 1) Explain why it is necessary to prevent any bubble of air from forming in the corked erlenmeyer flask when determining the volume of an irregular sample

It is necessary to prevent any bubble of air from forming while measuring the volume of an irregular sample because you have to ensure that your measurement of the volume is accurate as possible.

- 2) Suggest a method for measuring the density of an irregular sample of wood

For an irregular sample of wood, you can submerge the wood in water to determine the volume. First, you would record the mass of the wood. Then, after submerging it into the water you would record the displacement. The displacement, in this case, would be your v (volume variable). Then multiply the mass times the volume and calculate accordingly.

- 3) Will a cube of ice float in a beaker of pure ethyl alcohol? Or a beaker of pure water? Please explain

An ice cube won't float in a beaker of pure ethyl alcohol because it has more density, but it would float perfectly in pure water.

- 4) A supposedly gold nugget is tested to determine its density. It is found to displace 193 mL of water and has a mass of 2,371 grams. Determine if the nugget is made of gold or not and explain why or why not.

The density of gold is about 19.3g/cc. The nugget can't be made of gold based on the calculations shown below.

$$d = m/v$$

$$m=2371$$

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
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