

CLAIM

Answer the original question: How can you distinguish between a physical and a chemical change?

A physical change is a change in the appearance of a substance but there is no change in the identity of the substance. A chemical change involves a chemical reaction that produces an entirely new substance.

REASONING

Use scientific principles and knowledge that you have about the topic to explain why your evidence supports your claim.

In today's lab, I was able to distinguish between physical and chemical changes because of the evidence obtained during each activity. Evidence of a chemical change included: color change, odor, texture change, temperature change, or a gas was produced. The following activities included evidence of a chemical change: lighting a match, mixing vinegar and chalk, adding Alka-Seltzer tablets to water, combustion of a candle wick, and burning sugar.

Evidence of a physical change included: crushing, breaking, dissolving, a state of matter change, or any other change of appearance that did not result in a chemical reaction. The following activities produced evidence that indicated a physical change: crushing chalk, dissolving sugar, breaking Alka-Seltzer tablets, blowing bubbles in water, adding food coloring to water, and melting of candle wax.

AFTER GATHERING THEIR EXPERIMENTAL EVIDENCE AND WATCHING A SHORT VIDEO CLIP, STUDENTS WILL WRITE A REASONING PARAGRAPH THAT ANSWERS THE SCIENTIFIC QUESTION "HOW CAN YOU DISTINGUISH BETWEEN A PHYSICAL AND A CHEMICAL CHANGE"

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