

group (that is, the amide group), which is represented by the following resonance structures:



Because it is more difficult (i.e., it would take more energy) to twist a double bond than a single bond, the four atoms in the amide group become locked in the same plane (Figure 16.19). Figure 16.20 depicts the repeating amide group in a polypeptide chain.

On the basis of models and X-ray diffraction data, Pauling deduced that there are two common structures for protein molecules, called the α helix and the β -pleated sheet. The α -helical structure of a polypeptide chain is shown in Figure 16.21. The helix is stabilized by *intramolecular* hydrogen bonds between the NH and CO groups of the main chain, giving rise to an overall rodlike shape. The CO group of each amino acid is hydrogen-bonded to the NH group of the amino acid that is four residues away in the sequence. In this manner all the main-chain CO and NH groups take part in hydrogen bonding. X-ray studies have shown that the structure of a number of proteins, including myoglobin and hemoglobin, consist of α helices to a great extent.

The β -pleated structure gives the protein a sheet-like structure rather than the rodlike structure of the α helix. The polypeptide chain is almost fully extended, and each chain forms many *intermolecular* hydrogen bonds with adjacent chains. Figure 16.22 shows the two different types of β -pleated structures, called *parallel* and *antiparallel*. Silk molecules possess the antiparallel structure. Because its polypeptide chains are already in an extended form, silk lacks elasticity and extensibility, but it is quite strong due to the many intermolecular hydrogen bonds.

It is customary to divide protein structure into four levels of organization. The *primary structure* refers to the unique amino acid sequence of the polypeptide chain. The *secondary structure* includes those parts of the polypeptide chain that are stabilized by a regular pattern of hydrogen bonds between the CO and NH groups of the backbone (for example, the α helix or β -pleated sheet). The term *tertiary structure* applies to the three-dimensional structure stabilized by dispersion forces, hydrogen bonding, and other intermolecular forces. It differs from secondary structure because the amino acids taking part in these interactions may be far apart in the polypeptide chain.

A protein molecule may be made up of more than one polypeptide chain. Thus, in addition to the various interactions *within* a chain that give rise to the secondary and tertiary structures, we must also consider the interaction *between* chains. The

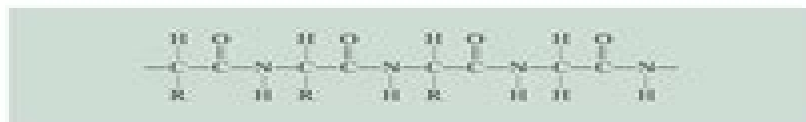


Figure 16.20 A polypeptide chain. Note the repeating units of the amide group. The symbol R represents part of the structure characteristic of the individual amino acids. For glycine, R is simply an H atom.

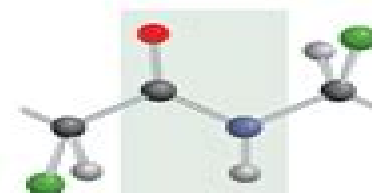


Figure 16.19 The planar amide group in proteins. Rotation about the peptide bond in the amide group is hindered by its double-bond character. The black atoms represent carbon; blue, nitrogen; red, oxygen; green, R groups; and gray, hydrogen.



Figure 16.21 The α -helical structure of a polypeptide chain. The structure is held in position by intramolecular hydrogen bonds, shown as dotted lines. For color key, see Figure 16.19.

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DJ Losen



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