

ABSTRACT

Extraction is the process of separating a substance from a mixture. In this experiment, we used an organic solvent (dichloromethane) to dissolve a carboxylic acid (benzoic acid) and a neutral organic compound, (p-dichlorobenzene). Since benzoic acid is insoluble in water, we had to find a way to make it soluble for the experiment to work. We used the extraction process as well as vacuum filtration to obtain the final results and calculations for this experiment.

Procedure

1. Into a stoppered separatory funnel add, 2g of a 1:1 mixture of benzoic acid and p-dichlorobenzene along with 30 ml of methylene chloride and swirl to dissolve.
 - a. Swirl and open the stopcock to release the pressure and continue doing so until no more gas is released.
2. Set up the equipment as shown below.

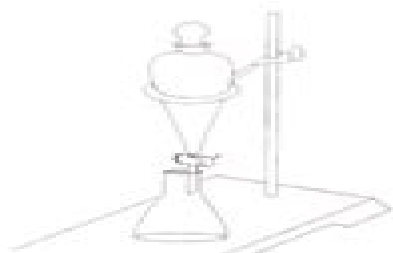


Figure 6.2 A separatory funnel properly supported, with an Erlenmeyer flask placed beneath the exit tube

3. Add 10 ml of sodium hydroxide through a shot stem funnel

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