



# Seminars In Alcohol And Drug Misuse College Seminars Series

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## **Seminars In Alcohol And Drug Misuse College Seminars Series:**

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House. Committee on Appropriations. Subcommittee on Department of Defense,2004      108-2 Hearings: Department of  
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web the enthalpy change of neutralisation is defined as the enthalpy change when solutions of an acid and alkali react together to produce one mole of water for example  $\text{HCl(aq)} + \text{NaOH(aq)} \rightarrow \text{NaCl(aq)} + \text{H}_2\text{O(l)}$  you can see that one mole of water is produced when one mole of hcl or one mole of naoh reacts