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Puja Mehta



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web ideal gas law google classroom a sample of H_2 gas is contained in a cylinder with a moveable piston at an initial pressure of P_1 the volume of the sample is increased from 3.0 L to 6.0 L while the temperature is held constant as shown in the diagram below

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web the observed behavior of gases embodied in the empirical gas laws leads to a series of equations that can be summarized by a single equation of state called the ideal gas law equation this shows the relationship between a gas's pressure P temperature T volume V and amount in moles n

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web each gas in a mixture of gases exerts its own pressure independently of the other gases present therefore the pressure of each gas within a mixture is called the partial pressure of the gas dalton's law of partial pressures states that the total pressure of a mixture of gases is equal to the sum of the partial pressures of the component gases

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web at very high pressures a gas will occupy a larger volume than predicted by the ideal gas law which of the following statements is true at low temperatures intermolecular forces become important and the pressure of a gas will be lower than predicted by