

## IBFT 2018 Methodical Practice/Packet for Chapter 7 and 8 (PLASMA)

### Multiple Choice

Simple formulae for the complex formulae is shown in parenthesis.

- \_\_\_\_\_ 1. The transmittance of a pure  $\text{Ca}^{2+}$  solution at a wavelength of 422 nm is 80%. The percent transmittance of a pure  $\text{Ca}^{2+}$  solution at a wavelength of 422 nm is 80%. The percent transmittance of a pure  $\text{Ca}^{2+}$  solution at a wavelength of 422 nm is 80%.
  - a. 10%
  - b. 20%
  - c. 60%
  - d. 80%
  
- \_\_\_\_\_ 2. The molar absorptivity of a substance is 1000 L/mol-cm. The molar absorptivity of a substance is 1000 L/mol-cm. The molar absorptivity of a substance is 1000 L/mol-cm.
  - a. 1000 L/mol-cm
  - b. 1000 L/mol-cm
  - c. 1000 L/mol-cm
  - d. 1000 L/mol-cm
  
- \_\_\_\_\_ 3. A solution of a substance has a molar absorptivity of 1000 L/mol-cm. The molar absorptivity of a substance is 1000 L/mol-cm. The molar absorptivity of a substance is 1000 L/mol-cm.
  - a. 1000 L/mol-cm
  - b. 1000 L/mol-cm
  - c. 1000 L/mol-cm
  - d. 1000 L/mol-cm
  
- \_\_\_\_\_ 4. A solution of a substance has a molar absorptivity of 1000 L/mol-cm. The molar absorptivity of a substance is 1000 L/mol-cm. The molar absorptivity of a substance is 1000 L/mol-cm.
  - a. 1000 L/mol-cm
  - b. 1000 L/mol-cm
  - c. 1000 L/mol-cm
  - d. 1000 L/mol-cm
  
- \_\_\_\_\_ 5. A solution of a substance has a molar absorptivity of 1000 L/mol-cm. The molar absorptivity of a substance is 1000 L/mol-cm. The molar absorptivity of a substance is 1000 L/mol-cm.
  - a. 1000 L/mol-cm
  - b. 1000 L/mol-cm
  - c. 1000 L/mol-cm
  - d. 1000 L/mol-cm

# Rspt Verbal Test 2 Pk1

**Denis Vincent, Mary Crumpler**



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