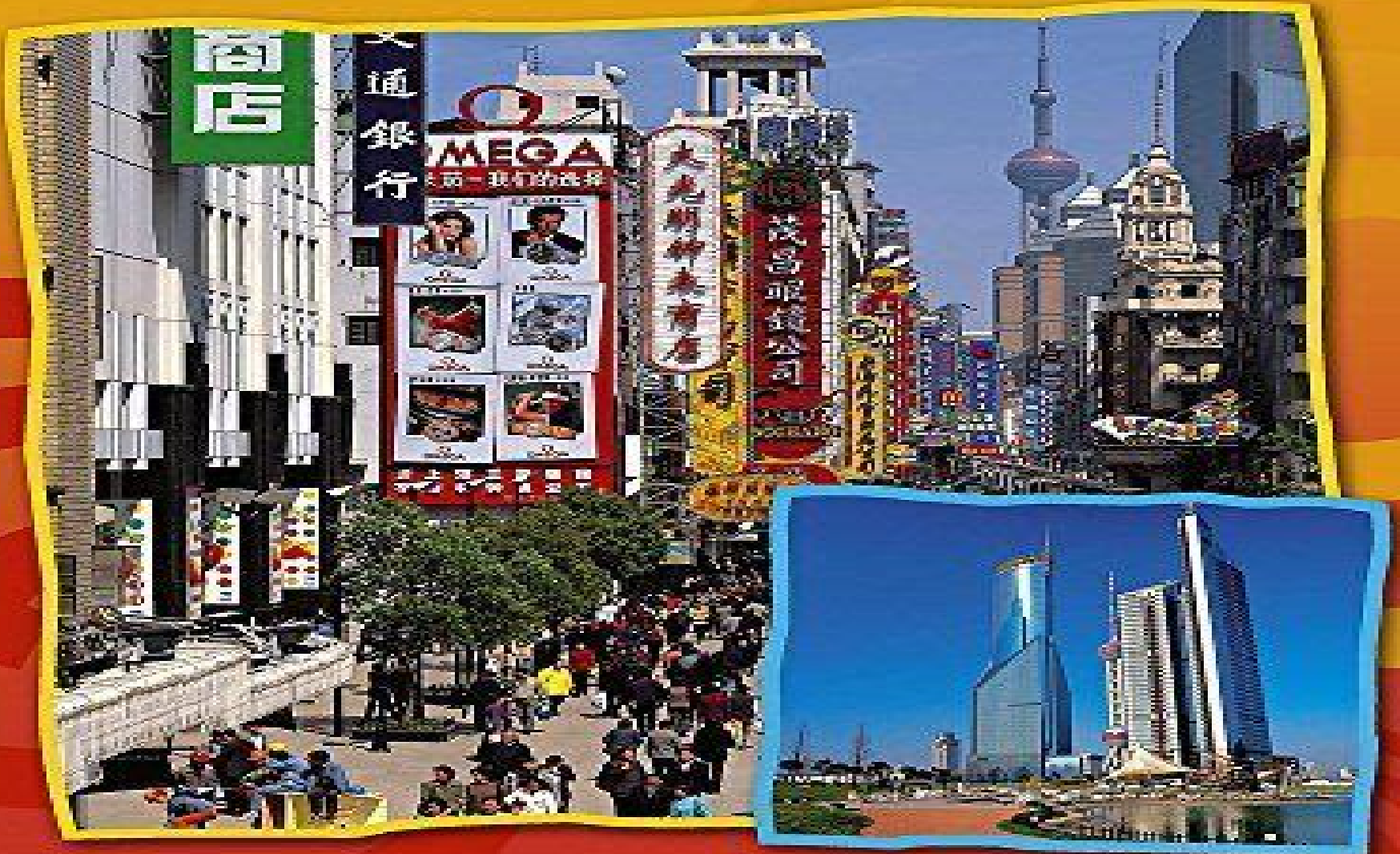


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Looking At China Looking At Other Countries

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web oct 13 2023 this means that you will need to balance the carbon atoms first 5 use a coefficient to balance the single carbon atom add a coefficient to the single carbon atom on the right of the equation to balance it with the 3 carbon atoms on the left of the equation $\text{C}_3\text{H}_8\text{O}_2 + 2\text{H}_2\text{O} \rightarrow 3\text{CO}_2$

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web solution a titrant volume 0 ml the solution ph is due to the acid ionization of hcl because this is a strong acid the ionization is complete and the hydronium ion molarity is 0 100 m the ph of the solution is then $\text{ph} = -\log 0.100 = 1.000$ b titrant volume 12.50 ml

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web expert answer 100 2 ratings the example is based upon henderson hasselbalch equation the equation gives relation between ph and pka of weak acid the equation is $\text{ph} = \text{pka} + \log \frac{[\text{salt}]}{[\text{acid}]}$ at equivalence point all a view the full answer transcribed image text

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theoretically and experimentally and to use an acid base titration to measure the effectiveness of antacid tablets summary a titration curve is the plot of the pH value of a solution vs the amount of titrant added

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