Radioisotope Techniques

Radioactivity is spontaneous disintegration of atomic nuclei. The nucleus emits α particles, β particles, or electromagnetic rays during this process. Approximately 3000 nuclides have been discovered so far that are unstable. They undergo spontaneous fission to achieve stability. During this process, the parent nuclide transforms into an atom of a different type, called the doughter nuclide.

Units of Radioactivity

Radioactivity is measured in cure. It is disintegration rate of 1 g radium that is 3.7×10^{19} disintegrations per second. SI unit for radioactivity is Becquerel (Bq), which is defined as one disintegration per second.

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1 Curie (Ci) = 3.7 \times 10^{10} disintegrations per second (dps) = 2.22 \times 10^{10} disintegrations per minute (dpm) 1 millicurie (mCi) = 3.7 \times 10^{7} dps 1 microcurie (\muCi) = 3.7 \times 10^{4} dps 1 Becquerel (Bq) = 1dps = 2.7 \times 10^{-11} Ci 1 Kilobecquerel (KBq) = 1dps = 2.7 \times 10^{-11} Ci
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F. E. McKinney

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four parts Part I deals with the fundamental concepts such as radioisotopes and labeled compounds the establishment and maintenance of a radioisotope laboratory and basic considerations in nuclear instrumentation Part II covers topics such as Geiger Muller and proportional counters semiconductor detectors and other systems for data accumulation and presentation Part III concerns itself with measurements of biological samples preparation of samples for liquid scintillation counting and involved equipment and radiochromatographic counting techniques Part IV tackles radioisotope measurements in vivo such as thyroid radioiodine uptake measurements single and multiple detector systems for whole body counting and large organic scintillation detectors The text is recommended for medical technologists and radiologists who would like to know more about the fundamentals applications and advances in the instrumentation involved in nuclear medicine NYO., 1960

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