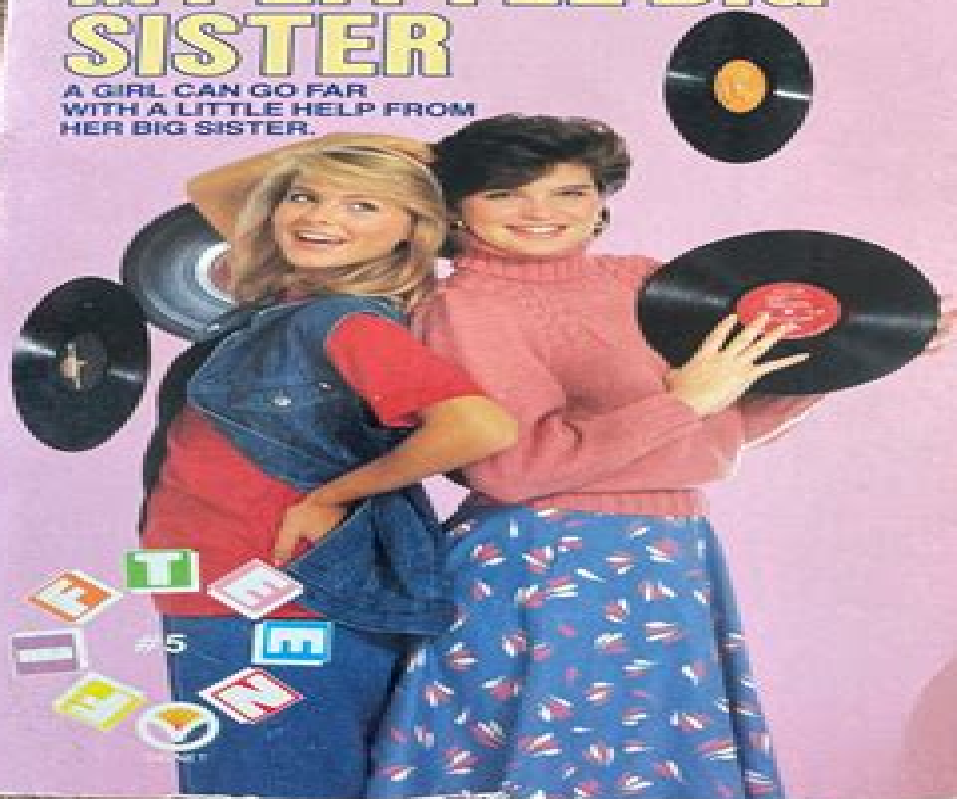


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mole moles of H_2O 40.1822 moles therefore according to the equation

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time the total pressure $p_a \times \text{mole fraction}_a = \text{mole}_a / \text{moles total}$ $\sum n_i$ all x_i n_i mole fraction of substance
a is often denoted by x_a and

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12 4 355 154 g mol 1

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