

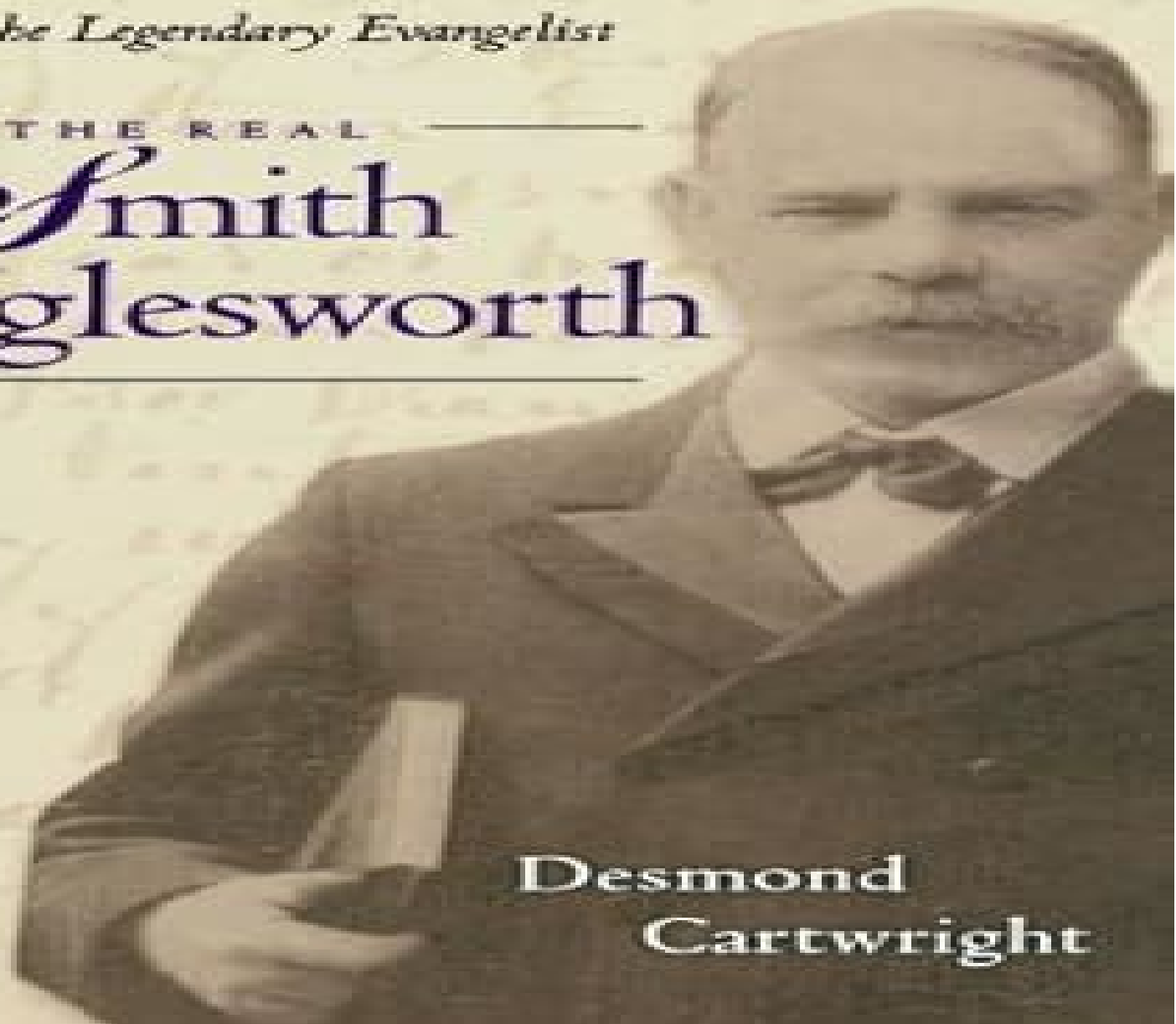
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writing a balanced nuclear equation  $^{137}_{55}\text{Cs} \rightarrow ^{137}_{56}\text{Ba} + ^0_0\text{e}$  because the mass number is the same on both sides of the equation  $x$  must have a mass number of 0

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remember that the two  $^1_0\text{n}$  neutrons  $^1_0\text{n}$  each contribute

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