

# MONTANA MAINSTREETS

VOLUME 6



## A GUIDE TO HISTORIC MISSOULA

BY ALLAN JAMES MATHEWS

# Montana Mainstreets Volume 6 A Guide To Historic Missoula

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## **Montana Mainstreets Volume 6 A Guide To Historic Missoula:**

*A Guide to Historic Missoula* Allan James Mathews, 2002 The sixth volume in the Montana Mainstreet series *A Guide to Historic Missoula* points readers to the buildings historic sites and parks that act as monuments to Missoula's and Montana's history

*Historic Underground Missoula* Nikki M. Manning, 2015-03-30 Much of Missoula's history lies beneath the surface As in many Old West cities cavernous underground tunnel systems purportedly hid countless nefarious activities from clandestine prostitution and Chinese opium dens to booze running during Prohibition These sordid tales captivate today's residents and beg questions about the city's furtive past Did local elite gentlemen mask their carnal habits there Did John Wayne really use the passageways to run personal errands unnoticed Author and urban archaeologist Nikki Manning ventures below to reconcile oral history with archaeological data in a fascinating exploration of Missoula's subterranean labyrinths

*Missoula* Stan B. Cohen, Svoboda Collection at the Northern Rockies Heritage Center, 2008-10-13 Missoula began in 1860 as Hellgate Village a trading post located about 4 miles west of what is now downtown In 1864 the construction of a mill powered by water diverted from Rattlesnake Creek drew residents of Hellgate to the new town and the arrival of the Northern Pacific Railroad in 1883 assured Missoula's future The establishment of the University of Montana led to growth on the south side of the Clark Fork River and the timber industry drove the economy Missoula has evolved into a trade medical educational and governmental center of western Montana

**Montana**, 2002

*A Guide to Historic Kalispell* Kathryn L. McKay, 2001 Founded in 1891 as a railroad town for the transcontinental Great Northern Railway Kalispell Montana faced bitter disappointment when the railroad relocated its route to the north in 1904 Most towns suffering similar fates fell into decline but not Kalispell which had become Flathead County seat in 1894 Situated in the majestic Northern Rockies the town thrived as the region's trade center and later became a gateway to Glacier National Park As Kalispell grew and prospered its streetscapes matured and the town's historic center remains a vibrant part of the community Discover Kalispell's rich history and architectural heritage through this lively guide 6 x 9 84 pages b w photos illustrations maps

**A Guide to Historic Hamilton** Chere Jiusto, 2000 Founded in 1890 against the backdrop of the lush Bitterroot Valley the town of Hamilton's history revolves around timber the railroad and agriculture Hamilton owes its early history to copper baron Marcus Daly whose Anaconda Company sawmill and private Bitter Root Stock Farm dominated the community through the late nineteenth century The drama of the twentieth century apple boom and the saga of the battle to cure Rocky Mountain tick fever enrich the town's more recent past Drawing the reader into the historical mosaic that is Hamilton Montana with architectural and historical information on town and valley history buildings and historic sites this guide is a treat for all those who would explore the jewel of the Bitterroot

**A Guide to Historic Lewistown** Ellen Sievert, Ken Sievert, 1999 In the Geographic Center of Montana sits Lewistown whose rich history is still reflected in today's streets A testament to the homesteading boom at the turn of the century Lewistown grew with the surrounding communities A service center rail stop

and county seat its population tripled between 1900 and 1910 then doubled again by 1920 It was not to last Drought and agricultural depression drove thousands of homesteaders from the land in the 1920s and 1930s Lewistown s boom had ended But preserved from that heady period of expansion are the exceptionally crafted business blocks homes churches often built of Lewistown red brick or local sandstone This architectural heritage awaits your discovery on the streets of Lewistown and through this lively guide **A Guide to Historic Glendive** ,1998 Young readers are invited along on an African safari

where they can see elephants spraying lions playing and hippos soaking **Montana Horse Racing** Brenda Wahler,2019-09-02 For centuries on prairie grasslands dusty streets and racing ovals everyday Montanans participated in the sport of kings More than a century after horses arrived in the region Lewis and Clark s Nez Perce guides staged horse races at Traveler s Rest in 1806 In response to hazardous street races the Montana legislature granted communities authority to ban immoderate riding or driving Helena led the way to respectable racing with Madam Coady s fashion course hosting the first territorial fair in 1868 Soon leading citizens like Marcus Daly built oval tracks and glitzy grandstands By 1890 a horse named Bob Wade set a world record for a quarter mile in Butte a mark that stood until 1958 Horsemanship and historian Brenda Wahler highlights the Big Sky s patrons of the turf and courageous equine champions including Kentucky Derby winner Spokane Hand Raised Chere Jiusto,Christine Brown,2011 Explore the hayloft stalls and hardware of a Montana barn and you will learn much about the state s farm and ranch traditions Crib barns with walls of timber stacked like Lincoln logs show the influence of French Canadian and Scandinavian immigrants Gambrel roofed barns which shed heavy snowfall and provide roomy haylofts tell of the long Montana winters that necessitated ample hay storage Tack rooms once filled with harnesses and gear tell of workhorses given shelter in heavy duty stalls nearby Beyond their utilitarian functions barns are simply beautiful Some stand proudly their freshly painted red lines contrasting sharply with the golden wheat in surrounding fields But some less fortunate are falling into disrepair Marked by rotting timbers and broken windowpanes these crumbling buildings still have much to teach us Historic Barns of Montana presents the best most unique most significant and most beautiful of these barns Photographer Tom Ferris explored barns inside and out across Montana snapping the hundreds of photographs in the book Authors and architectural historians Chere Jiusto and Christine Brown help readers understand the significance of what they are looking at and tell the stories of individual barns Historic Barns of Montana recognizes these buildings as both useful and beautiful encourages their preservation and honors the ranch and farm families that built them

A Guide to Historic Virginia City Marilyn Grant,1998 The gold rush era town of Virginia City recently purchased by the State of Montana to preserve for posterity makes a fitting first subject for the Montana Mainstreet series Once it was Montana s acting territorial capital and the center of trade for Alder Gulch the site of the richest placer mines in the world but Virginia City became a town almost frozen in time once gold deposits played out and the state capital moved to Helena in 1889 Today Virginia City attracts visitors from all over the world who marvel at its intact architecture If walking down

Virginia City's streets is like a trip backwards in time the road map for that journey is Guide to Historic Virginia City  
Forthcoming Books Rose Arny,2003      **American Book Publishing Record** ,2003      *The Bicentennial of the United States of America* American Revolution Bicentennial Administration,1977      News for Members - Minnesota Historical Society Minnesota Historical Society,1946-11      **The Genealogical Helper** ,1982      **Federal Register** ,1977      Kliatt Young Adult Paperback Book Guide ,1998      *Leahy's Hotel-motel Guide and Travel Atlas of the United States, Canada, and Mexico* ,1952      *Spartacus International Gay Guide* ,2006

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web identify the changes in oxidation states and write the oxidation half reaction and the reduction half reaction and for each half reaction balance the atom undergoing redox changes if necessary add the number of electrons that correspond to the change in oxidation state

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web 1 define each remember oil rig oxidation is loss of e reduction is gain of e a oxidation b reduction c oxidizing agent d reducing agent 2 write half reactions for each of the following atoms or ions label each as oxidation or reduction use the activity series chart a al b ba2 c br 2 d ca e ga3 f h 2 g h

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web during oxidation and reduction the oxidation numbers of atoms change if an atom is oxidized its oxidation number increases ie it becomes more ve or less ve if an atom is reduced its oxidation number decreases ie it becomes less ve or more ve

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web you should try to answer the questions without referring to your textbook if you get stuck try asking another group for help 1 identify the oxidation state of each element in the following a na 2 so 4 b mno 4 c kclo 3 d ch 3 oh 2 identify the oxidation state of nitrogen in the following a no 3 b no 2 c no 2

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web an oxidation reduction redox reaction is one in which one or more atoms change oxidation numbers oxidation occurs when an atom s oxidation state becomes more positive indicating that electrons have been lost reduction occurs when an atom s oxidation state becomes less positive indicating that electrons have been gained



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web oxidation reduction reactions commonly known as redox reactions are reactions that involve the transfer of electrons from one species to another the species that loses electrons is said to be oxidized while the species that gains electrons is said to be reduced

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web describe redox reactions in terms of electron transfer recall that oxidation involves loss of electrons and reduction involves gain of electrons represent simple redox reactions using half equations identify oxidising and reducing agents for simple redox reactions understand the reactivity of metals and metal displacement reaction

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web what is an oxidation reduction or redox reaction this is a reaction in which electrons are transferred between reactants what do the following terms mean oxidation oxidation is a loss of electrons reduction reduction is a gain of electrons what is a helpful way to remember this

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web write the oxidation and reduction reactions for each redox reaction the first one is done for you 23  $\text{Fe}^{2+} + \text{Co} \rightarrow \text{Co}^{2+} + \text{Fe}$   
oxidation  $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^{-}$  reduction  $\text{Fe}^{2+} + 2\text{e}^{-} \rightarrow \text{Fe}$  24 3  $\text{Ag} + \text{Ni}^{3+} \rightarrow \text{Ni} + 3\text{Ag}^{+}$  oxidation reduction 25  $\text{Cu}^{2+} + \text{Pb} \rightarrow \text{Pb}^{2+} + \text{Cu}$  oxidation reduction 26  $\text{O}_2 + 2\text{H}_2 \rightarrow 2\text{H}_2\text{O}$

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web in each of the following redox reactions indicate the elements that undergo changes in oxidation number and the magnitude of change then identify the element oxidized and reduced as well as the oxidizing agent and reducing agent a  $\text{I}_2 + 5\text{CO} \rightarrow \text{I}_2\text{O}_5 + 5\text{CO}_2$  b  $\text{C} + \text{O}_2 \rightarrow \text{CO}_2$  c  $\text{Fe} + \text{O}_2 \rightarrow \text{Fe}_2\text{O}_3$  d  $\text{Fe} + \text{H}_2\text{SO}_4 \rightarrow \text{Fe}_2(\text{SO}_4)_3 + \text{H}_2$

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web chemical reactions in which there is a transfer of electrons from one substance to another are known as oxidation reduction reactions or redox reactions in this unit you will examine the oxidation reduction process and use the oxidation

state and oxidation number concepts to identify redox reactions and to keep track of electrons

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web jun 28 2010 answers i determine what is oxidized and what in each reaction identify the oxidizing agent and the reducing agent also  $2\text{Sr} + \text{O}_2 \rightarrow 2\text{SrO}$   $\text{Sr}^0$  to  $\text{Sr}^2$  oxidized reducing agent  $\text{O}^0$  to  $\text{O}^2$  reduced ox ag  $2\text{Li} + \text{S} \rightarrow \text{Li}_2\text{S}$   $\text{Li}^0$  to  $\text{Li}^1$  oxidized red ag  $\text{S}^0$  to  $\text{S}^2$  reduced ox ag  $3\text{C} + 2\text{CS}_2 \rightarrow \text{CS}_2 + 2\text{C}$   $\text{C}^0$  to  $\text{C}^2$  oxidized reducing agent  $\text{S}^0$  to  $\text{S}^2$  reduced ox ag