

# SCANDINAVIAN

DENMARK/NORWAY/SWEDEN

*cooking for pleasure* / Jytte Hardisty



# Scandinavian Cooking For Pleasure Denmark Norway Sweden By Hardisty Jytte

**Mark Steyvers**



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40 g of  $\text{H}_2\text{O}$  molar mass of  $\text{H}_2\text{O}$  is 18 gm mole and molar mass of  $\text{CH}_3\text{OH}$  is 32 gm mole solution moles of  $\text{CH}_3\text{OH}$  4.532 0.14 mole moles of  $\text{H}_2\text{O}$  40.18 2.2 moles therefore according to the equation

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web notice that the mole fraction has no units on it and is written as a decimal value do not change it to percent note of caution you could see the term mole percent it is simply the mole fraction multiplied by 100 for example in the problem just below the mole fraction of cinnamic acid is 0.2885 its mole percent would be 28.85

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