# AIR, COMBUSTION, RUSTING AND FIRE 17.A sample of iron rusts. Explain w

- Extinguishing a fire caused by burning kerosene, carbon dioxide is used in preference to water. Explain why?
- Give a reason why, carbon dioxide is used as a fire extinguisher?
- Carbon dioxide is not used as a fire extinguisher in class D fires. Explain why?
- Burning magnesium continue to burn inside the gas jar full of carbon dioxide. Explain why?
- 5. A piece of burning magnesium continues to burn in a gas jar of Sulphur dioxide gas. Explain why?
- A small child has been playing with matches and its clothes have caught fire. Explain how you would put out the flames.
- Zinc is used to cover iron so as to prevent rusting of iron. Explain why?
- 8. Explain why motor vehicles based in Mombasa rust much faster than those based in Nairobi?
- If the tin surface is scratched then the iron will rust. Why does this not happen with zinc plating (galvanizing)?
- 10. Aluminium metal would protect iron from rusting better than zinc metal. Explain why?
- 11.It is not advisable to wash vehicles using sea water, Explain why?
- 12. Why do car exhaust rust faster than other structures made of steel?
- 13. The parts of shipwrecks made of iron rust more slowly in deep water. Explain why?
- 14.Alkaline medium inhibits the rusting of iron. Explain why?
- 15. Iron does not rust even if zinc coating is broken in a galvanized iron pipe. Explain why?
- 16.Zinc is used to coat for iron and not vice versa. Explain why?

- 17.A sample of iron weighs more after it rusts. Explain why?
  - 18. The percent of carbon dioxide in the atmosphere does not significantly vary from 0.04%. Explain why?
- 19. The air over an open flame is always moist. Explain why?
- 20.Explain why, cooking pots made up of aluminium do not corrode easily when exposed to air?
- 21. Why does a magnesium ribbon be cleaned before burning in air?
- 22. When a candle was burnt completely, the total m ass of the products was found to be greater than the original mass of the candle. Explain why?
- 23.In temperate countries, salt is sprayed on the roads to defrost and clear roads but the long-term effect on this practice is costly to motorists. Explain why?
- 24. Why baking soda does extinguish fire?
- 25. Tetrachloromethane is used as a fire extinguisher but not carbon disulphide. Explain why?
- 26.Explain why magnesium gains in mass but coal loses mass when each is burned in air?

#### HEAT SOURCES AND FALAMES

- when air holes of a Bunsen burner are fully opened the Bunsen-burner produce a non- luminous flame. Explain why?
- It is advisable to set the Bunsen burner to luminous flame prior to experiment. Explain why?
- it is advisable to turn a non-luminous flame to luminous flame while the flame is not in use. Explain why?
- 4. Non -luminous flame is used for heating. Explain why by giving two reasons?
- Explain how hotness of Bunsen burner flame can be increased.

## **Reasoning In Chemistry Pb 1995**

**Georgios Tsaparlis** 

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Chemical Education: Towards Research-based Practice J.K. Gilbert, Onno de Jong, Rosária Justi, David F. Treagust, Jan H. van Driel, 2006-03-11 Chemical education is essential to everybody because it deals with ideas that play major roles in personal social and economic decisions This book is based on three principles that all aspects of chemical education should be associated with research that the development of opportunities for chemical education should be both a continuous process and be linked to research and that the professional development of all those associated with chemical education should make extensive and diverse use of that research It is intended for pre service and practising chemistry teachers and lecturers chemistry teacher educators chemical education researchers the designers and managers of formal chemical curricula informal chemical educators authors of textbooks and curriculum support materials practising chemists and chemical technologists It addresses the relation between chemistry and chemical education curricula for chemical education teaching and learning about chemical compounds and chemical change the development of teachers the development of chemical education as a field of enquiry This is mainly done in respect of the full range of formal education contexts schools universities vocational colleges but also in respect of informal education contexts books science centres and museums

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strengthen existing connections between educational researchers and practitioners and 3 highlight how people have used chemical education based research in their teaching practice There are places in the literature dedicated for chemical education research CER however there is not a clear avenue for those that have changed their teaching methods based on published CER and report their experiences Creating this article collection will foster collaboration between chemical education researchers and teachers of organic chemistry. This opportunity allows these instructors to share evidence based practices experiences challenges and innovative approaches from CER literature and beyond This Research Topic bridges discipline based education research and the scholarship of teaching and learning which will help advance organic chemistry education and improve student outcomes Problems and Problem Solving in Chemistry Education Georgies Tsaparlis, 2021-05-17 Problem solving is central to the teaching and learning of chemistry at secondary tertiary and post tertiary levels of education opening to students and professional chemists alike a whole new world for analysing data looking for patterns and making deductions As an important higher order thinking skill problem solving also constitutes a major research field in science education Relevant education research is an ongoing process with recent developments occurring not only in the area of quantitative computational problems but also in qualitative problem solving The following situations are considered some general others with a focus on specific areas of chemistry quantitative problems qualitative reasoning metacognition and resource activation deconstructing the problem solving process an overview of the working memory hypothesis reasoning with the electron pushing formalism scaffolding organic synthesis skills spectroscopy for structural characterization in organic chemistry enzyme kinetics problem solving in the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner **Cumulated Index Medicus** ,1995 An Introduction to Thermal Physics Daniel **Chemistry** Exam Leaders Expert, V. Schroeder, 2021 This is a textbook for the standard undergraduate level course in thermal physics sometimes called thermodynamics or statistical mechanics Originally published in 1999 it quickly gained market share and has now been the most widely used English language text for such courses as taught in physics departments for more than a decade Its clear and accessible writing style has also made it popular among graduate students and professionals who want to gain abetter understanding of thermal physics The book explores applications to engineering chemistry biology geology atmospheric science astrophysics cosmology and everyday life It includes two appendices reference data an annotated bibliography a complete index and 486 homework problems Roald Hoffmann on the Philosophy, Art, and Science of Chemistry Jeffrey Kovac, Michael Weisberg, 2012-02-01 Over a career that spans nearly fifty years Roald Hoffmann has thought and written copiously about the broader context of chemistry and its relationship to philosophy and poetry This book gathers

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