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Making FRONTPAGE® Work: Essential Tips & Techniques

Konrad King

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web apr 22 2020 molar volume of gases 1 counting in moles one mole is defined as 6 10 23 particles this is the avogadro number an unchanging constant whether you are counting hydrogen atoms carbon dioxide molecules or carbonate ions a mole is always 6 10 23 but as different particles have different mass their molar mass varies

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web zero degrees celsius or 273~k is defined as standard temperature and one atmosphere or 760~mm hg is defined as standard pressure the volume occupied by one mole of a gas is called molar volume therefore the molar volume of a gas at standard temperature and pressure stp is 22~4~l

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