



Chem 110B Practice Final Exam

Part A: Short Answer/Calculation Questions

1. What energy difference separates these two species? Is the following calculation correct?
 $9.4788 \text{ g} - 9.4688 \text{ g} = 0.0100 \text{ g}$

eq 1 eq 2 eq 3 eq 4 eq 5 _____

2. What energy difference exists between F^{2+} and F^{3+} ?

eq 6 eq 7 eq 8 eq 9 eq 10 _____

3. What energy difference exists between F^{2+} and F^{3+} ?

eq 11 or 12 eq 13 or 14 eq 15 or 16 eq 17 or 18 _____

4. What are the following reactions? The product is H_2O .

eq 19 or 20 eq 21 or 22 eq 23 or 24 _____

5. What volume of gas at 1 atm and 25°C is required to produce the reaction of 100 g of H_2O at 25°C?

eq 25 or 26 eq 27 or 28 eq 29 or 30 eq 31 or 32 _____

6. Complete the following table by supplying the chemical formulae of the compounds.

Species	Formula
	Fe_2O_3
Calcium phosphate	
	Fe_2O_3
	Calcium

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