



i) calculating molar masses (the mass of 1 mol)

Calculate the molar mass of the following substances.

1) 1 mol of O atoms = $\boxed{16 \text{ g O/mol}}$

2) 1 mol of Oxygen gas (O_2) = $16 \text{ g} \times 2 = \boxed{32 \text{ g O}_2/\text{mol}}$

3) 1 mol of Ozone gas (O_3) = $16 \text{ g} \times 3 = \boxed{48 \text{ g O}_3/\text{mol}}$

4) 1 mol of Nitric Acid HNO_3 $\left\{ \begin{array}{l} \text{H} = 1 \times 1 = 1 \\ \text{N} = 1 \times 14 = 14 \\ \text{O} = 3 \times 16 = 48 \end{array} \right\} \boxed{63 \text{ g HNO}_3/\text{mol}}$

5) 1 mol $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ $\left\{ \begin{array}{l} \text{Cu} = 64 = 64 \\ \text{S} = 32 \times 1 = 32 \\ \text{O} = 16 \times 4 = 64 \\ \text{H}_2\text{O} = 18 \times 5 = 90 \end{array} \right\} \boxed{250 \text{ g CuSO}_4 \cdot 5\text{H}_2\text{O}/\text{mol}}$



Mole Notes

K Morrison



Mole Notes:

Lecture Notes on Fundamentals of Combustion , Notes On Thermodynamics: Hot Oolong Cools Hway Chuan Kang,2024-06-11 This title is a supplement to lectures and tutorials in a Thermodynamics course and also serves as a guide to more comprehensive texts Rather than merely hurrying through the principles and then dealing with applications the book presents the scientific method by discussing the science of thermodynamics starting from empirical observations that are relatable to students For example the book uses everyday experiences such as a cup of hot tea cooling spontaneously to arrive at the Second Law and Entropy through the idea of the heat engine All the fundamentals are covered and illustrated with examples that resonate with the broad concerns and interests of students who take STEM classes today The book examines the thermodynamics of hydrogen and gasoline engines fuel cells versus the explosive combustion of hydrogen how efficiently organisms and Spiderman utilize energy the fizzing of a can of soda and decompression sickness and how atmospheric carbon dioxide affects ocean pH and worryingly dissolves the calcium carbonate shells of marine animals and also what might happen if you inadvertently fall into a salt lake Thermodynamics is presented as the macroscopic approach to understanding Nature when heat is involved The book draws upon the idea of microstates where that clarifies the macroscopic ideas entropy of mixing of gases is linked to Boltzmann and Gibbs entropy formulations thus motivating the formulation of the chemical potential of non ideal systems in terms of their activities Thermodynamics contains deep insight into the passage of Time In the discussions of the Second Law the book highlights this emphasizing that all the processes we observe in our universe are irreversible Adopting an informal and readable style without compromising the rigour in this book the goal is to help a broad audience of students appreciate the essential meaning of the Laws of Thermodynamics and to apply the fundamental framework at an elementary level Notes and Queries ,1916 *NEET UG Physics Study Notes with Theory + Practice MCQs for Complete Preparation | Based on New Syllabus as per NMC* EduGorilla Prep Experts,2023-12-01 Nature Notes ,1892 Mole Notes Michael Benedikt,1971 **NEET UG Chemistry Paper Study Notes |Chapter Wise Note Book For NEET Aspirants | Complete Preparation Guide with Self Assessment Exercise** EduGorilla Prep Experts,2022-09-15 Best Selling Book in English Edition for NEET UG Chemistry Paper Exam with objective type questions as per the latest syllabus Increase your chances of selection by 16X NEET UG Chemistry Paper Study Notes Kit comes with well structured Content Chapter wise Practice Tests for your self evaluation Clear exam with good grades using thoroughly Researched Content by experts **The Selborne Magazine and "Nature Notes"** ,1913 Properties of Solutions - Quick Review Chemistry Notes and Outline E Staff, Properties of Solutions Quick Review Outline and Handout Learn and review on the go Use Quick Review Chemistry Notes to help you learn or brush up on the subject quickly You can use the review notes as a reference to understand the subject better and improve your grades Easy to remember facts to help you perform better Perfect study notes for all high school and college students 9 Pages NBS Technical Note ,1973-08

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