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genuine love and respect
for Armstrong's art. . . .
This is likely to be the
standard biography of
Armstrong for some time."

— *Washington Post*



Louis ARMSTRONG

AN EXTRAVAGANT LIFE

LAURENCE
BERGREEN

Author of As Thousands Cheer: The Life of Irving Berlin

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 $\text{hcl aq} + \text{naoh aq} \rightarrow \text{nacl aq} + \text{h}_2\text{O l}$ one mole of hcl would be fully neutralized by one mole of naoh

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being naoh reaction with a 2nd substance $\text{H}_2\text{C}_2\text{O}_4 + 2\text{NaOH} \rightarrow \text{Na}_2\text{C}_2\text{O}_4 + 2\text{H}_2\text{O}$ to differentiate an acid from a base in a solution a pH scale is used the pH scale ranges from 0 to 14 a pH of 7 is neutral in an acid base reaction

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web this point of titration is referred to as the equivalence point considering the 1 1 stoichiometry of this acid base reaction
 $\text{naoh aq} + \text{c}_6\text{h}_4\text{cooh cook aq} \rightarrow \text{c}_6\text{h}_4\text{coona cook aq} + \text{h}_2\text{o l}$
the point of equivalence is the point of titration when the number of moles of naoh na added is equal to the number of moles of khp nb in the

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