

* Thermodynamics: Branch of science; deals with the concepts of heat & temperature & the inter-conversion of heat & other forms of energy.

* Equation of state of an ideal gas:

for n moles of an ideal gas.

$$PV = nRT$$

* Thermodynamic Equilibrium: A system is said to be in the state of thermodynamic equilibrium if the macroscopic variables describing the thermodynamic state of the system do not change with time.

* Thermal Equilibrium: Two systems are in thermal equilibrium with each other if they have same temperature.

* Thermodynamic Variables or Parameters:

i) Extensive Variables: These variables depend on the mass of the system or the number of constituent particles.

ii) Intensive Variables: These variables are independent of the mass or the number of constituent particles of the system.

eg. i) Volume, Internal Energy, entropy.

ii) specific heat, surface tension, density.

Jyoshika Saka

Notes Of Thermodynamics Statistics

J Dewey



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