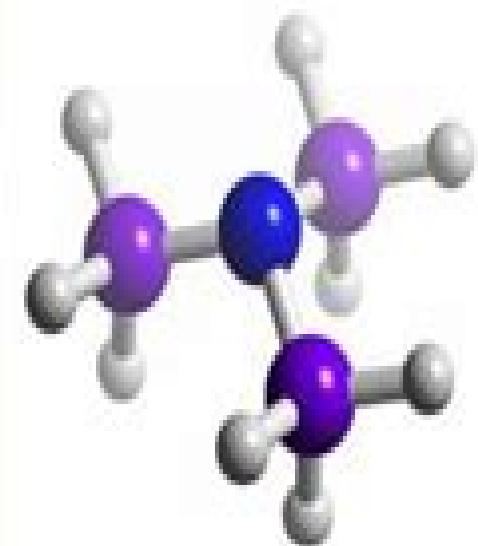


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Química general Gordon M. Barrow, 1975-11 Sin introducir grandes modificaciones en los temas que habitualmente se encuentran en un texto de esta clase el autor ha tratado de mostrar que la Química se origina para proporcionar una perspectiva adecuada y entrar en el tema *Química general. Introducción a la Química Teórica*.

Cristóbal Valenzuela Calahorro, 1995-01-01 **Química general** Ralph S. Becker, Wayne E. Wentworth, 1977 Esta Química general ha sido escrita para un curso completo de análisis pensando en alumnos de Ciencias e Ingeniería. Este libro es riguroso en el sentido de que es completo. Al mismo tiempo se ha escrito a un nivel tal que pueda ser comprendido por los alumnos que inician sus estudios universitarios. No presupone conocimientos de Cálculo matemático ni de Química previamente adquiridos para su estudio. *Química General Manual de Prácticas de Laboratorio* Instituto Tecnológico de Santo Domingo,

Resolución de problemas de química general R. Nelson Smith, Conway Pierce, 1991 Este libro sirve a aquellos alumnos que empiezan a estudiar Química y tienen dificultades al analizar los problemas y encontrar soluciones lógicas o que tienen dificultades con las representaciones gráficas y su interpretación.

Química general Kenneth W. Whitten, Kenneth D. Gailey, 1985 Ácidos bases y sales. Cinética química. Compuestos de coordinación. Disoluciones. Electroquímica. Elementos no metálicos. Enlace covalente y nomenclatura inorgánica. Equilibrio químico. Equilibrios en disolución acuosa. Estequiométria. Símbolos químicos. Fórmulas y ecuaciones. Estructura atómica. Los gases y la teoría cinetomolecular. Líquidos y sólidos. Metales en transición. Metales y metalurgia. Orbitales moleculares y enlace químico. Periodicidad química y enlace iónico. Química nuclear. Química orgánica. Reacciones de oxidación-reducción. Reacciones químicas. Reacciones químicas: estudio sistemático.

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Resolución de problemas de química general Christopher J. Willis, 1980 Este libro acerca de la resolución de problemas intenta ser un complemento de los textos elementales de Química. Incluye la mayor parte de los temas que se consideran durante un curso y resulta aconsejable tanto para estudiar bajo la dirección del profesor como para hacerlo el alumno por su propia cuenta o con un mínimo de supervisión.

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