



Reaktionen in der Organischen Chemie 1

Alkane

Che

*nucleophile Substitutionen



primäre Substrate



sekundäre Substrate



Radikale

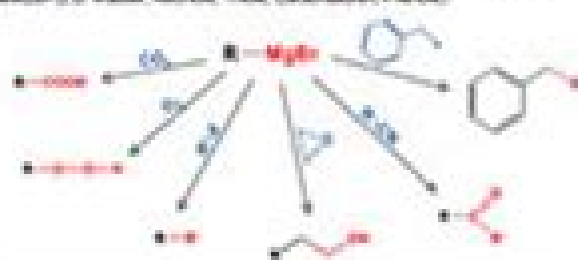
Radikale, die mindestens ein freies Elektronenpaar haben und oft negativ geladen sind (anionisch) bzw. eine starke negative Partialladung besitzen. Sie treten unter Verwendung eines Radikalengenerators wie katalytische Wirkung in einem geschlossenen System auf.

Substitution

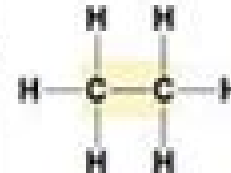
Radikalische

Reaktionen von Grignard-Verbindungen

Grignard-Verbindungen reagieren mit Substraten, welche einen Protonenakzeptor (z.B. Wasser, Alkohole, Thiole, Carbonsäuren, Ester).



Wichtig: Die radikalische Substitution findet bevorzugt an sekundären und tertiären Kohlenstoff-Atomen statt (abhängig von der Substitution der Kohlenstoff-Atome).

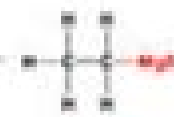


die Lichtquelle ist für die Bildung von Radikalen wichtig

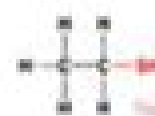


LiAlH₄

Grignard-Synthese



Mercaptan-Synthese*



Hydroxy-Substitution*



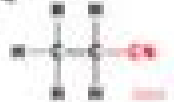
Finkelstein-Reaktion* (halogenatom-Vertauschung)



Ammonolyse*



Kolbe-Nitrilsynthese*



Halogenierung (radikalische Substitution)



Wurtz-Reaktion



Eliminierung



Williamson-Ethersynthese*



Oxidation

(Verbrennung)

Die radikalische Substitution funktioniert auch mit 1,1- und 1,2-Dibrom-Verbindungen, die in der Reaktion umgewandelt werden können.

Verbindungen, die in der Reaktion umgewandelt werden können.

Reaktionsmechanismen dieser Stoffgruppen sind auf einer anderen Übersicht dargestellt.

Radikale Und Radikationen In Der Organischen Synthese

Camilla Rothe



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