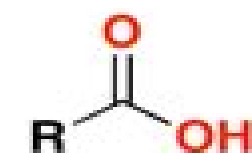
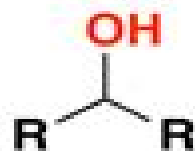


Once deciphered, chromic acid is a fairly straightforward reagent. It oxidizes primary alcohols to carboxylic acids and secondary alcohols to ketones. It will also oxidize aldehydes to ketones

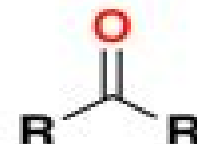
*Starting material*



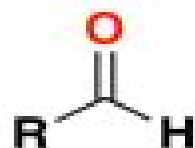
*Carboxylic acid*



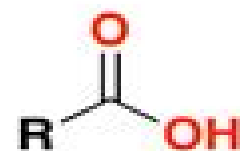
*Secondary alcohol*



*Ketone*



*Aldehyde*



*Carboxylic acid*

# Oxidation Hazards More Than Just Air

**CH Cherryholmes**



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