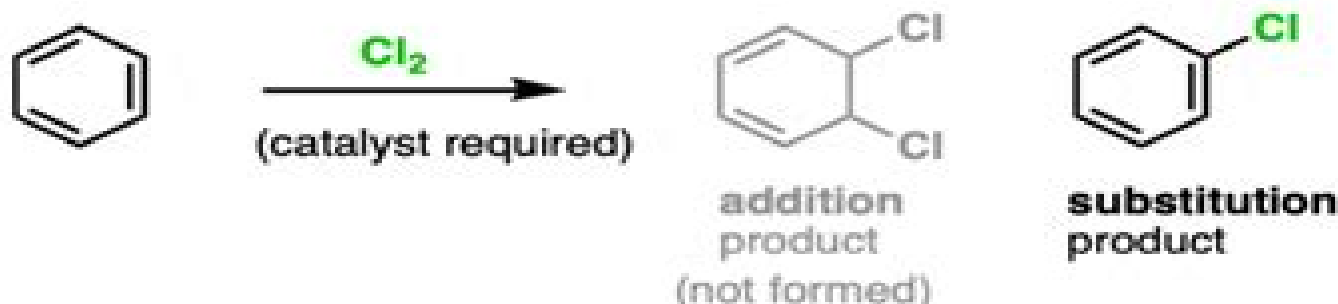


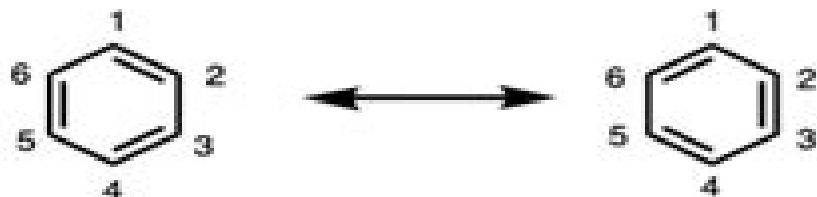
Summary: Aromaticity

Aromaticity is the term given to describe the unusual **stability** of benzene and related molecules. Not related to smells! 😊

- Benzene does not easily undergo **addition** like normal alkenes
- When benzene does undergo reaction, it tends to undergo **substitution** reactions
(and even then, often requires heating and addition of a catalyst)

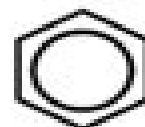


- The measured **resonance energy** of benzene is extremely high (36 kcal/mol) compared to conjugated alkenes like 1,3-cyclohexadiene (2 kcal/mol)
- Unlike dienes, which have alternating bond lengths, **all C-C bond lengths in benzene are identical** (1.40 Å)



Equivalent resonance structures
(not an equilibrium!)

Either structure for benzene is correct
and they can be used interchangeably



Hybrid of these two forms
sometimes drawn like this
to show that the pi-electrons
are "**delocalized**"

- The property of aromaticity is not unique to benzene

Physical Chemistry Of 12dithiole Compounds The Question Of Aromaticity

David Baud



Physical Chemistry Of 12dithiole Compounds The Question Of Aromaticity:

Unveiling the Magic of Words: A Overview of "**Physical Chemistry Of 12dithiole Compounds The Question Of Aromaticity**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Physical Chemistry Of 12dithiole Compounds The Question Of Aromaticity**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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