



Chapter - 12

Organic Chemistry : Some Basic Principles and Techniques

FAST TRACK : QUICK REVISION

ORGANIC CHEMISTRY

It deals with the study of hydrocarbons (compounds of carbon and hydrogen elements) and their derivatives.

Some organic compounds may also contain nitrogen, oxygen, sulphur, phosphorus, halogens, etc.

Berzelius, proposed that a 'vital force' was responsible for the formation of organic compounds.

This was rejected by F. Wohler who synthesised first organic compound urea from an inorganic compound.



- Acetic acid was synthesised by Kolbe and methane by Berthelot.
- Types of hybridisation of C-atom :

Hybridisation	Structure	Bond angle	Examples
sp^3	Tetrahedral	$109^\circ 28'$	Ethane, Methane
sp^2	Trigonal	120°	Ethene, Propene
sp	Linear	180°	Ethyne, Propyne

- Reasons for existence of large number of organic compounds:
- **Catenation** : The property of atoms of an element to link with one another forming chains of identical atoms is called *catenation*. Carbon exhibits catenation to the maximum extent.
- **Isomerism** : It is the property by virtue of which two or more compounds have the same molecular formula but different physical or chemical properties.

Physical Methods Of Organic Chemistry Techniques Of Organic Chemistry S

Christian Drosten



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