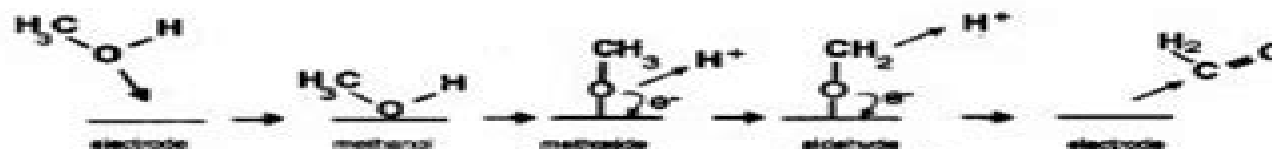


reactions especially pathways leading to H_2CO have been unexplored until recently due to interferences which inhibit detection of the intermediate species



Scheme 1: Mechanism suggested for methanol oxidation to H_2CO [31].

by modern in situ analysis techniques like mass spectrometry [35,36] and infrared spectroscopy [37-39]. Direct detection of H_2CO by mass spectrometry is generally hindered by the methanol mass fragments at $m/z = 28-30$ [27, 35, 36]. IR spectroscopy is also incapable of producing H_2CO spectra as H_2CO in aqueous solution forms a gem-diol ($\text{H}_2\text{C}(\text{OH})_2$) which overlaps with water vibrational bands [37-39]. Recent studies [27, 28, 30, 40] have brought serious attention to the stable by-products of methanol oxidation, as these affect CO_2 production and decrease energy conversion efficiency [9, 41-43]. These efforts mainly involved the use of differential electrochemical mass spectrometry (DEMS), which can only detect CO_2 and methyl formate directly. In their DEMS studies, assuming that CO_2 , formic acid and H_2CO are the only direct reaction products, the faradic current for H_2CO was found indirectly as the difference between the total Faradic charges measured and the partial currents/charges of CO_2 and formic acid formation, calculated using corresponding calibration constants [27, 28, 40]. In this research project, we are mainly interested in

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