

A zero order homogeneous gas phase reaction with stoichiometry,  $A \rightarrow rR$ , proceeds in a constant volume bomb,  $\pi = 1$  when  $t = 0$  and  $\pi = 1.5$  when  $t = 1$ . If the same reaction, same feed composition and initial pressure proceeds in a constant pressure setup, find  $V$  at  $t = 1$  if  $V = 1$  at  $t = 0$ . ( $\pi$  = total pressure;  $t$  = time,  $V$  = volume.)

6

# Reaction Kinetics Volume 1 Homogeneous Gas

**Shasha Hu**



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