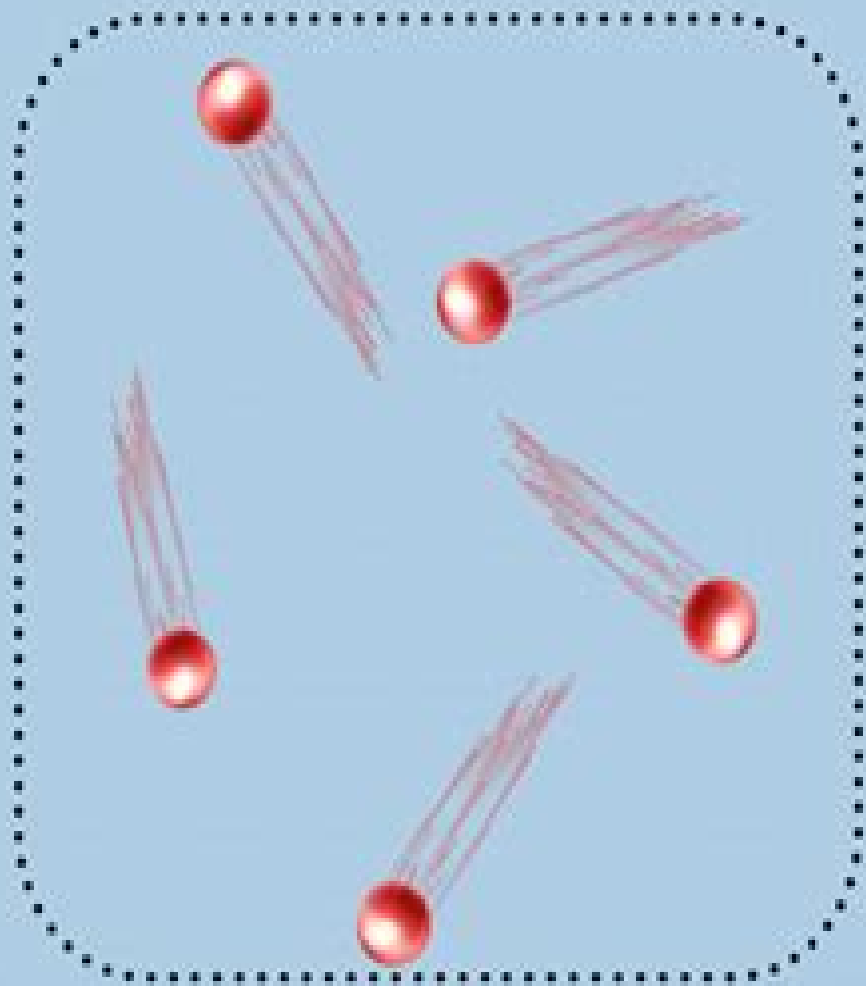


# Kinetic Molecular Theory of Gases

- Many gas particles, so statistics applies
- Particles are tiny and spread far apart
- Gas particles are in constant random motion
- No attraction or repulsion between particles
- Collisions are elastic



# Molecular Theory Of Gases And Liquids

**Paul van der Schoot**



## **Molecular Theory Of Gases And Liquids:**

The Molecular Theory of Gases and Liquids Joseph O. Hirschfelder, Charles F. Curtiss, R. Byron Bird, 1964-01-15 An essential cross disciplinary reference for molecular interactions Molecular Theory of Gases and Liquids offers a rigorous comprehensive treatment of molecular characteristics and behaviors in the gaseous and fluid states A unique cross disciplinary approach provides useful insight for students of chemistry chemical engineering fluid dynamics and a variety of related fields with thorough derivations and in depth explanations throughout Appropriate for graduate students and working scientists alike this book details advanced concepts without sacrificing depth of coverage or technical detail *Molecular Theory Of Gases And Liquids* Joseph O. Hirschfelder, 1981 Molecular Theory of Gases and Liquids Joseph O. Hirschfelder, 1964 Molecular Theory of Gases & Liquids Joseph O. Hirschfelder, 1954 Molecular theory of gases and liquids Robert Byron Bird, Joseph Oakland Hirschfelder, Charles Francis Curtiss, 1954 Molecular Theory of Gases and Liquids Joseph Oakland Hirschfelder, 2003-01 Molecular Theory of Gases and Liquids Curtiss CF., 1954 **Molecular Theory of Gases and Liquids [by] Joseph O. Hirschfelder, Charles F. Curtiss [and] R. Byron Bird, with the Assistance of the Staff of the Former University of Wisconsin Naval Research Laboratory** Joseph O. Hirschfelder, 1964 A Kinetic Theory of Gases and Liquids Richard Daniel Kleeman, 1920 Molecular Theory of Gases and Liquids. (Stichworte Teil 3) Joseph Oakland Hirschfelder, Robert Byron Bird, Charles Francis Curtiss, 1954 **Molecular Theory of Gases and Liquids**, 1967 *A Kinetic Theory of Gases and Liquids* Richard D. Kleeman, 2022-10-27 This work has been selected by scholars as being culturally important and is part of the knowledge base of civilization as we know it This work is in the public domain in the United States of America and possibly other nations Within the United States you may freely copy and distribute this work as no entity individual or corporate has a copyright on the body of the work Scholars believe and we concur that this work is important enough to be preserved reproduced and made generally available to the public We appreciate your support of the preservation process and thank you for being an important part of keeping this knowledge alive and relevant *Molecular Theory of Nematic (and Other) Liquid Crystals* Paul van der Schoot, 2022-07-29 This book provides a didactic derivation of the main theories of thermotropic and lyotropic liquid crystals revealing the common molecular theoretic framework that underpins both theories This unified context will help young researchers in coming to grips with the basics of the simplest of liquid crystals being uniaxial nematic liquid crystals easing them into the intricacies of more complex forms of such materials irrespective of whether they are thermotropic or lyotropic The coverage provides a theoretical understanding of the phase behaviour that is what drives molecules and particles to spontaneously align themselves as well as an appreciation of the role of entropy energy and so on The focus here is on the main theories for the isotropic nematic transition being the Maier Saupe and the Onsager theories and how they are derived from a common description known as classical density functional theory DFT This book will be a valuable resource for senior undergraduate

and graduate students and experimentalists and engineers who feel intimidated by more formal or rigorous theoretical accounts and textbooks Exercises at the end of each chapter help the reader to apply the basic concepts also to other types of liquid crystal in particular the smectic liquid crystal

*Molecular Theory of Gases and Liquids. [By] J. O. Hirschfelder ... Charles F. Curtiss ... R. Byron Bird, etc* Joseph Oakland Hirschfelder, Robert Byron Bird, Charles Francis Curtiss, 1954

**A Kinetic Theory of Gases and Liquids (Classic Reprint)** Richard D. Kleeman, 2017-12 Excerpt from A Kinetic Theory of Gases and Liquids In constructing a general Kinetic Theory the problem that presents itself first for investigation is the dependence of the velocity of translation of a molecule in a substance on its density and temperature It is often assumed that this velocity is the same in the liquid as in the gaseous state at the same temperature It can be shown however that this holds only for each molecule at the instant it passes through a point in the substance at which the forces of the surrounding molecules neutralize each other The total average velocity corresponding to the whole path of a molecule is usually much greater than the foregoing velocity in a liquid and dense gas on account of the effect of the molecular forces of attraction and repulsion About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books Find more at [www.forgottenbooks.com](http://www.forgottenbooks.com) This book is a reproduction of an important historical work Forgotten Books uses state of the art technology to digitally reconstruct the work preserving the original format whilst repairing imperfections present in the aged copy In rare cases an imperfection in the original such as a blemish or missing page may be replicated in our edition We do however repair the vast majority of imperfections successfully any imperfections that remain are intentionally left to preserve the state of such historical works

**A Kinetic Theory of Gases and Liquids** Richard Daniel Kleeman, 2013-09 This historic book may have numerous typos and missing text Purchasers can usually download a free scanned copy of the original book without typos from the publisher Not indexed Not illustrated 1920 edition Excerpt of gas from which the molecule comes and thus increases with decrease of pressure The difference between  $V_a$  and  $V_c$  therefore also increases with decrease of pressure which increases the value of the ratio  $T_j$   $i_j$  according to the foregoing equation This corresponds to a decrease of and thus the effect of slipping becomes the greater the lower the pressure The dynamical mechanism underlying the result that the viscosity of a gas is independent of its density may be GASEOUS VISCOSITY AND THE TEMPERATURE

127 illustrated by the following considerations A molecule transfers a certain amount of momentum to the gas at the end of each transfer distance on migrating at right angles to the motion of the gas in the direction of the decrease of motion If the concentration of the gas is halved the length of each transfer distance is doubled while the momentum transferred at the end of each transfer distance is also doubled since the velocity gradient of the gas remains the same Since a change in molecular concentration of a gas does not alter the molecular velocities the momentum transferred per second by a molecule moving between two parallel plates of material one of which is at rest while the other moves parallel to itself is in the latter case double that in the former But since the number of molecules per cubic cm available for momentum transference in the

former case is half the number in the latter the total momentum transferred is in each case the same or the viscosity of the gas has not been altered by altering its density The quantity  $K$  is a function of the temperature This is shown by the calculated values of  $K$  contained in Table X for a number of gases at different temperatures These values are not corrected for

Physical Chemistry for the Biosciences Raymond Chang, 2005-02-11 This book is ideal for use in a one semester introductory course in physical chemistry for students of life sciences The author's aim is to emphasize the understanding of physical concepts rather than focus on precise mathematical development or on actual experimental details Subsequently only basic skills of differential and integral calculus are required for understanding the equations The end of chapter problems have both physiochemical and biological applications

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