

Molecular structure model

(2019 version) (10/10/2019 & 10/10/2019)

Atom	Ball weight	Color	Size (mm)	Qty
H (Hydrogen)	1.0	White	10	100
C (Carbon)	12.0	Black	20	100
O (Oxygen)	16.0	Red	20	100
N (Nitrogen)	14.0	Blue	20	100
F (Fluorine)	18.0	Green	20	100
Cl (Chlorine)	35.5	Yellow	30	100
S (Sulfur)	32.1	Brown	30	100
P (Phosphorus)	31.0	Purple	30	100
Br (Bromine)	79.9	Dark Green	40	100
I (Iodine)	126.9	Dark Purple	50	100
Li (Lithium)	7.0	Grey	10	100
Na (Sodium)	23.0	Grey	15	100
K (Potassium)	39.1	Grey	20	100
Ca (Calcium)	40.1	Grey	25	100
Mg (Magnesium)	24.3	Grey	20	100
Al (Aluminum)	27.0	Grey	25	100
Si (Silicon)	28.1	Grey	25	100
B (Boron)	10.8	Grey	15	100
Cu (Copper)	63.5	Grey	30	100
Zn (Zinc)	65.4	Grey	30	100
Fe (Iron)	55.8	Grey	30	100
Co (Cobalt)	58.9	Grey	30	100
Ni (Nickel)	58.7	Grey	30	100
Pd (Palladium)	106.4	Grey	40	100
Ag (Silver)	107.9	Grey	40	100
Au (Gold)	197.0	Grey	50	100
Pl (Platinum)	195.1	Grey	50	100
Os (Osmium)	190.2	Grey	50	100
Ir (Iridium)	192.2	Grey	50	100
Rh (Rhodium)	102.9	Grey	30	100
Sn (Tin)	118.7	Grey	40	100
Pb (Lead)	207.2	Grey	60	100
Ba (Barium)	137.3	Grey	40	100
La (Lanthanum)	138.9	Grey	40	100
Ce (Cerium)	140.1	Grey	40	100
Pr (Praseodymium)	140.9	Grey	40	100
Nd (Neodymium)	144.2	Grey	40	100
Pm (Promethium)	145.0	Grey	40	100
Sm (Samarium)	150.4	Grey	40	100
Eu (Europium)	152.0	Grey	40	100
Gd (Gadolinium)	157.3	Grey	40	100
Tb (Terbium)	158.9	Grey	40	100
Dy (Dysprosium)	162.5	Grey	40	100
Ho (Holmium)	164.9	Grey	40	100
Er (Erbium)	167.3	Grey	40	100
Tm (Thulium)	168.9	Grey	40	100
Yb (Ytterbium)	173.1	Grey	40	100
Lu (Lutetium)	175.0	Grey	40	100
Hf (Hafnium)	178.5	Grey	40	100
Ta (Tantalum)	180.9	Grey	40	100
W (Tungsten)	183.8	Grey	40	100
Re (Rhenium)	186.2	Grey	40	100
Os (Osmium)	190.2	Grey	40	100
Ir (Iridium)	192.2	Grey	40	100
Pt (Platinum)	195.1	Grey	40	100
Au (Gold)	197.0	Grey	40	100
Hg (Mercury)	200.6	Grey	40	100
Tl (Thallium)	204.4	Grey	40	100
Pb (Lead)	207.2	Grey	40	100
Bi (Bismuth)	209.0	Grey	40	100
Po (Polonium)	209.0	Grey	40	100
At (Astatine)	210.0	Grey	40	100
Rn (Radon)	222.0	Grey	40	100
Fr (Francium)	223.0	Grey	40	100
Ra (Radium)	226.0	Grey	40	100
Ac (Actinium)	227.0	Grey	40	100
Th (Thorium)	232.0	Grey	40	100
Pa (Protactinium)	231.0	Grey	40	100
U (Uranium)	238.0	Grey	40	100
Np (Neptunium)	237.0	Grey	40	100
Pu (Plutonium)	244.0	Grey	40	100
Am (Americium)	243.0	Grey	40	100
Cm (Curium)	247.0	Grey	40	100
Bk (Berkelium)	247.0	Grey	40	100
Cf (Californium)	251.0	Grey	40	100
Es (Einsteinium)	252.0	Grey	40	100
Fm (Fermium)	257.0	Grey	40	100
Md (Mendelevium)	258.0	Grey	40	100
No (Nobelium)	259.0	Grey	40	100
Lr (Lawrencium)	262.0	Grey	40	100

Here are a few simple models



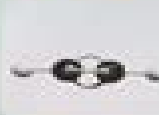
H₂O



CO₂



CH₄



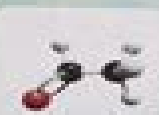
NH₃



C₂H₆



H₂O₂



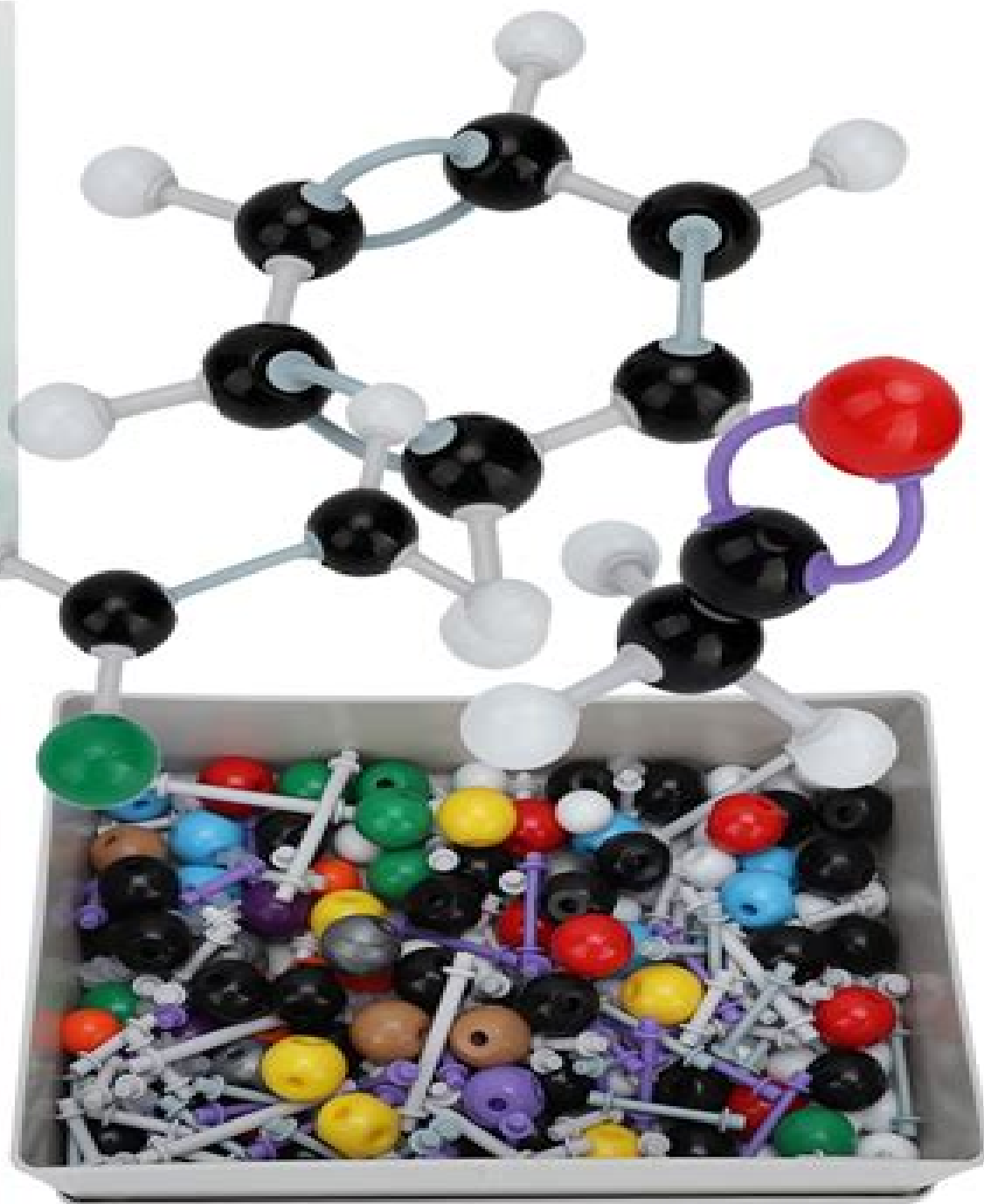
HCN



CO



CO₂



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PT Brinkman



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Molecular Orbital Calculations for Biological Systems Anne-Marie Sapse, 1998-11-12 Molecular Orbital Calculations for Biological Systems is a hands on guide to computational quantum chemistry and its applications in organic chemistry biochemistry and molecular biology With improvements in software molecular modeling techniques are now becoming widely available they are increasingly used to complement experimental results saving significant amounts of lab time Common applications include pharmaceutical research and development for example ab initio and semi empirical methods are playing important roles in peptide investigations and in drug design The opening chapters provide an introduction for the non quantum chemist to the basic quantum chemistry methods ab initio semi empirical and density functionals as well as to one of the main families of computer programs the Gaussian series The second part then describes current research which applies quantum chemistry methods to such biological systems as amino acids peptides and anti cancer drugs Throughout the authors seek to encourage biochemists to discover aspects of their own research which might benefit from computational work They also show that the methods are accessible to researchers from a wide range of mathematical backgrounds Combining concise introductions with practical advice this volume will be an invaluable tool for research on biological systems The Prentice Hall Molecular Model Set for Organic Chemistry, 1983 This kit enables users to build virtually all simple molecules encountered in organic chemistry Includes space filling models that simulate the true shape of saturated compounds Provides open models that form realistic single double and triple bonds even strained rings Allows smooth rotation of the bonds to make conformational analysis easy Contains enough components to create several models at once The components are precision tooled from quality plastics are virtually indestructible and come in a sturdy plastic case for easy storage Provides a useful Instruction Book with photos diagrams and concise discussions of chemical principles

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