

ADSORPTION

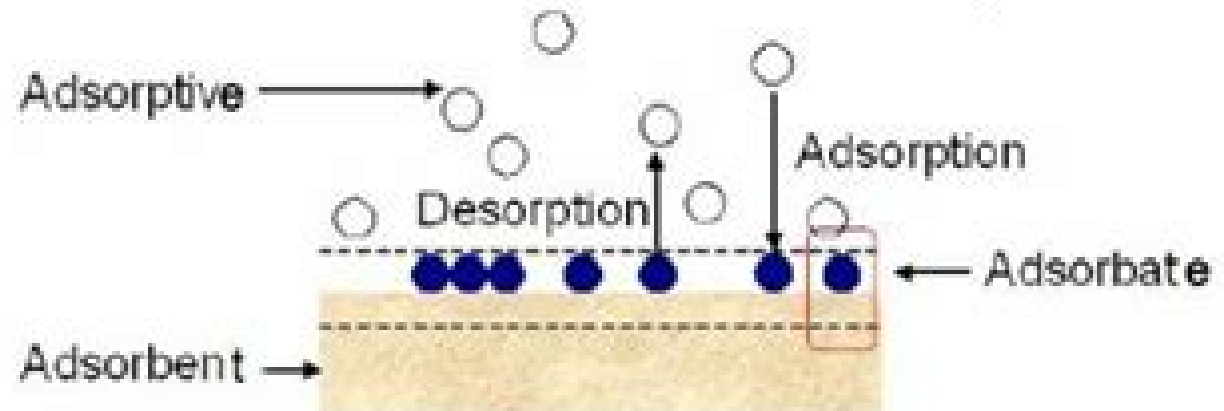
Adsorption is the process in which matter is extracted from one phase and concentrated at the surface of a second phase. (Interface accumulation). This is a surface phenomenon as opposed to absorption where matter changes solution phase, e.g. gas transfer. This is demonstrated in the following schematic.

If we have to remove soluble material from the solution phase, but the material is neither volatile nor biodegradable, we often employ adsorption processes.

adsorbate: material being adsorbed

adsorbent: material doing the adsorbing. (examples are activated carbon or ion exchange resin).

In a typical adsorption process, a gaseous, liquid or solid molecule (the *adsorptive*) attached to a solid or liquid surface, the *adsorbent*, and forms the *adsorbate*, a complex between *adsorptive* and *adsorbent*.



Principles Of Adsorption And Adsorption Processes

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