HOEKSTRA: Relative to the isotopic ratio of standard sea water as determined by Epstein.

The effect of impurities can be illustrated by the Colorado Plateau samples (Shinarump No. 1 claim) of 65 and 85% purity. A considerable change in O¹⁸ content was brought about by the final purification of the sample. The Canadian samples (Nos. 9-12) are less pure than we would have liked, and they show a considerable variation in O¹⁸ content from sample to sample.

Results on the synthetic pitchblende obtained from Professor Gruner would suggest that at 200°C very little, if any, oxygen isotope fractionation occurs, since the observed O¹⁸ content is about what one would expect from tap water at Minneapolis.

DOLE: I don't understand how the oxygen-uranium ratio could be 2.33 and the synthetic pitchblende 98% pure.

HOEKSTRA: This is merely the ratio of the molecular weights of the oxides. Professor Gruner prepared these pitchblendes by the reduction of uranyl sulfate in solution. It is evident that his synthetic pitchblende was not completely reduced, that precipitation can occur without complete reduction of the uranyl ion. Beginning at the other extreme, a similar oxygen-uranium ratio may be obtained when tetravalent uranium is precipitated in contact with air at room temperature, or even at the boiling point of water; the freshly precipitated oxide reacts rapidly with oxygen of the air to an oxygen-uranium ratio of 2.33 or higher. Complete oxidation to UO₃ can occur. The usual oxidation product of uranium dioxide, U₃O₈, is not formed.

KEITH: Would you mind saying again how these were made?

HOEKSTRA: Professor Gruner prepared these compounds by heating, in a sealed tube, a uranyl sulfate solution saturated with hydrogen sulfide in the presence of a certain amount of ferrous ion.

KEITH: Would you expect under those conditions to get a relationship between the isotope ratios and the temperature?

HOEKSTRA: You would not expect much fractionation, but as Silverman found with silica, fractionation does occur in some systems even at quite high temperatures--certainly a measurable effect.

CRAIG: We find this to be true for calcium carbonates, but apparently there is not much fractionation here.

HOEKSTRA: The effect of weathering is best shown in the Spruce Pine samples which are highly oxidized.

We noted that many of the pitchblendes seemed to have an isotopic composition very similar to that of ground waters in the region where they were found. This induced us to go back and check oxygen exchange between pitchblendes and

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