

\* Thermodynamics: Branch of science; deals with the concepts of heat & temperature & the inter-conversion of heat & other forms of energy.

\* Equation of state of an ideal gas:

for  $n$  moles of an ideal gas.

$$PV = nRT$$

\* Thermodynamic Equilibrium: A system is said to be in the state of thermodynamic equilibrium if the macroscopic variables describing the thermodynamic state of the system do not change with time.

\* Thermal Equilibrium: Two systems are in thermal equilibrium with each other if they have same temperature.

\* Thermodynamic Variables or Parameters:

i) Extensive Variables: These variables depend on the mass of the system or the number of constituent particles.

ii) Intensive Variables: These variables are independent of the mass or the number of constituent particles of the system.

eg. i) Volume, Internal Energy, entropy.

ii) specific heat, surface tension, density.

Jyoshika Saka

# Notes Of Thermodynamics Statistics

**Jessica J Manson**



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