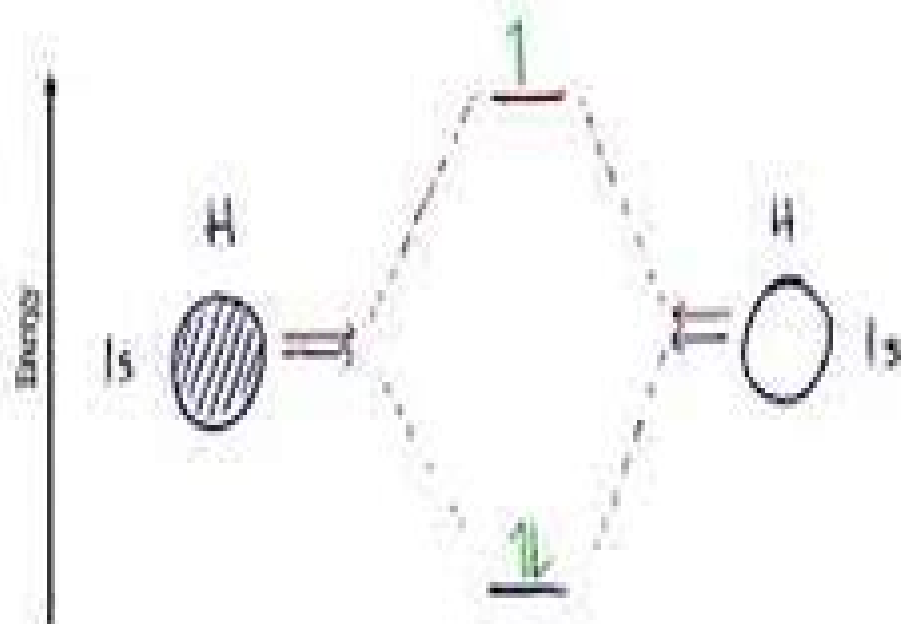


Molecular Orbital Theory

1. In-phase overlap $\rightarrow$ bonding orbital, low in energy.
2. Out-of-phase overlap $\rightarrow$ antibonding orbital, high in energy.
3. Fill orbitals with electrons.



3 valence electrons

Molecular Orbital Theory For Organic Chemists

JR Anderson

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