

AIR, COMBUSTION, RUSTING AND FIRE FIGHTING

1. Extinguishing a fire caused by burning kerosene, carbon dioxide is used in preference to water. Explain why?
2. Give a reason why, carbon dioxide is used as a fire extinguisher?
3. Carbon dioxide is not used as a fire extinguisher in class D fires. Explain why?
4. Burning magnesium continue to burn inside the gas jar full of carbon dioxide. Explain why?
5. A piece of burning magnesium continues to burn in a gas jar of Sulphur dioxide gas. Explain why?
6. A small child has been playing with matches and its clothes have caught fire. Explain how you would put out the flames.
7. Zinc is used to cover iron so as to prevent rusting of iron. Explain why?
8. Explain why motor vehicles based in Mombasa rust much faster than those based in Nairobi?
9. If the tin surface is scratched then the iron will rust. Why does this not happen with zinc plating (galvanizing)?
10. Aluminium metal would protect iron from rusting better than zinc metal. Explain why?
11. It is not advisable to wash vehicles using sea water. Explain why?
12. Why do car exhaust rust faster than other structures made of steel?
13. The parts of shipwrecks made of iron rust more slowly in deep water. Explain why?
14. Alkaline medium inhibits the rusting of iron. Explain why?
15. Iron does not rust even if zinc coating is broken in a galvanized iron pipe. Explain why?
16. Zinc is used to coat for iron and not vice versa. Explain why?

17. A sample of iron weighs more after it rusts. Explain why?
18. The percent of carbon dioxide in the atmosphere does not significantly vary from 0.04%. Explain why?
19. The air over an open flame is always moist. Explain why?
20. Explain why, cooking pots made up of aluminium do not corrode easily when exposed to air?
21. Why does a magnesium ribbon be cleaned before burning in air?
22. When a candle was burnt completely, the total mass of the products was found to be greater than the original mass of the candle. Explain why?
23. In temperate countries, salt is sprayed on the roads to defrost and clear roads but the long-term effect on this practice is costly to motorists. Explain why?
24. Why baking soda does extinguish fire?
25. Tetrachloromethane is used as a fire extinguisher but not carbon disulphide. Explain why?
26. Explain why magnesium gains in mass but coal loses mass when each is burned in air?

HEAT SOURCES AND FLAMES

1. When air holes of a Bunsen burner are fully opened the Bunsen-burner produce a non-luminous flame. Explain why?
2. It is advisable to set the Bunsen burner to luminous flame prior to experiment. Explain why?
3. It is advisable to turn a non-luminous flame to luminous flame while the flame is not in use. Explain why?
4. Non-luminous flame is used for heating. Explain why by giving two reasons?
5. Explain how hotness of Bunsen burner flame can be increased.

Reasoning In Chemistry Pb 1995

Puja Mehta



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This book delves into Reasoning In Chemistry Pb 1995. Reasoning In Chemistry Pb 1995 is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Reasoning In Chemistry Pb 1995, encompassing both the fundamentals and more intricate discussions.

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- Chapter 3: Reasoning In Chemistry Pb 1995 in Everyday Life
- Chapter 4: Reasoning In Chemistry Pb 1995 in Specific Contexts
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