

# MODERN PHYSICAL CHEMISTRY

An Introductory Text

H. Block

A. H. Holliday

Butterworths

# Modern Physical Chemistry An Introductory Text

**Robert G. Mortimer**



### **Modern Physical Chemistry An Introductory Text:**

**Modern physical chemistry** Hermann Block, Arthur K. Holliday, 1976      **Fundamentals of Physical Chemistry [by] H.d.crockford & Samuel B.knight** H.D. Crockford, 1967      **Modern Physical Chemistry: an Introduction Text** Hermann Block, Arthur Kenneth Holliday, 1973      **Modern Physical Chemistry** G.H. Duffey, 2013-11-11 In this new textbook on physical chemistry fundamentals are introduced simply yet in more depth than is common Topics are arranged in a progressive pattern with simpler theory early and more complicated theory later General principles are induced from key experimental results Some mathematical background is supplied where it would be helpful Each chapter includes worked out examples and numerous references Extensive problems review and discussion questions are included for each chapter More detail than is common is devoted to the nature of work and heat and how they differ Introductory Caratheodory theory and the standard integrating factor for dGrev are carefully developed The fundamental role played by uncertainty and symmetry in quantum mechanics is emphasized In chemical kinetics various methods for determined rate laws are presented The key mechanisms are detailed Considerable statistical mechanics and reaction rate theory are then surveyed Professor Duffey has given us a most readable easily followed text in physical chemistry      **Physical Chemistry** William M. Davis, 2011-12-06 Designed for a two semester introductory course sequence in physical chemistry Physical Chemistry A Modern Introduction Second Edition offers a streamlined introduction to the subject Focusing on core concepts the text stresses fundamental issues and includes basic examples rather than the myriad of applications often presented in other more encyclopedic books Physical chemistry need not appear as a large assortment of different disconnected and sometimes intimidating topics Instead students should see that physical chemistry provides a coherent framework for chemical knowledge from the molecular to the macroscopic level The book offers Novel organization to foster student understanding giving students the strongest sophistication in the least amount of time and preparing them to tackle more challenging topics Strong problem solving emphasis with numerous end of chapter practice exercises over two dozen in text worked examples and a number of clearly identified spreadsheet exercises A quick review in calculus via an appendix providing the necessary mathematical background for the study of physical chemistry Powerful streamlined development of group theory and advanced topics in quantum mechanics via appendices covering molecular symmetry and special quantum mechanical approaches      *Kinetic Theory of Gases* Walter Kauzmann, 2013-04-22 This monograph and text was designed for first year students of physical chemistry who require further details of kinetic theory The treatment focuses chiefly on the molecular basis of important thermodynamic properties of gases including pressure temperature and thermal energy Includes numerous exercises many partially worked out and end of chapter problems 1966 edition      *TEXTBOOK OF PHYSICAL CHEMISTRY* H. K. MOUDGIL, 2014-10-21 This comprehensive textbook now in its second edition is mainly written as per the latest syllabi of physical chemistry of all the leading universities of India as well as the new syllabus recommended by the UGC This

thoroughly revised and updated edition covers the principal areas of physical chemistry such as thermodynamics quantum chemistry molecular spectroscopy chemical kinetics electrochemistry and nanotechnology In a methodical and accessible style the book discusses classical irreversible and statistical thermodynamics and statistical mechanics and describes macroscopic chemical systems steady states and thermodynamics at a molecular level It elaborates the underlying principles of quantum mechanics molecular spectroscopy X ray crystallography and solid state chemistry along with their applications The book explains various instrumentation techniques such as potentiometry polarography voltammetry conductometry and coulometry It also describes kinetics rate laws and chemical processes at the electrodes In addition the text deals with chemistry of corrosion and nanomaterials This text is primarily designed for the undergraduate and postgraduate students of chemistry B Sc and M Sc for their course in physical chemistry Key Features Gives a thorough treatment to ensure a solid grasp of the material Presents a large number of figures and diagrams that help amplify key concepts Contains several worked out examples for better understanding of the subject matter Provides numerous chapter end exercises to foster conceptual understanding

*A Textbook of Physical Chemistry* Arthur Adamson, 2012-12-02 A Textbook of Physical Chemistry Second Edition serves as an introductory text to physical chemistry Topics covered range from wave mechanics and chemical bonding to molecular spectroscopy and photochemistry ideal and nonideal gases the three laws of thermodynamics thermochemistry and solutions of nonelectrolytes The kinetics of gas phase reactions colloids and macromolecules and nuclear chemistry and radiochemistry are also discussed This edition is comprised of 22 chapters the first of which introduces the reader to the behavior of ideal and nonideal gases with particular emphasis on the van der Waals equation The discussion then turns to the kinetic molecular theory of gases and the application of the Boltzmann principle to the treatment of molar polarization dipole and magnetic moments the phenomenology of light absorption and classical and statistical thermodynamics The chapters that follow focus on the traditional sequence of chemical and phase equilibria electrochemistry and chemical kinetics in gas phase and solution phase This book also considers wave mechanics and its applications molecular spectroscopy and photochemistry and the excited state and then concludes with an analysis of crystal structure colloid and polymer chemistry and radio and nuclear chemistry This reference material is intended primarily as an introductory text for students of physical chemistry

**Physical Chemistry** Robert G. Mortimer, 2000-04-28 This new edition of Robert G Mortimer's Physical Chemistry has been thoroughly revised for use in a full year course in modern physical chemistry In this edition Mortimer has included recent developments in the theories of chemical reaction kinetics and molecular quantum mechanics as well as in the experimental study of extremely rapid chemical reactions While Mortimer has made substantial improvements in the selection and updating of topics he has retained the clarity of presentation the integration of description and theory and the level of rigor that made the first edition so successful Emphasizes clarity every aspect of the first edition has been examined and revised as needed to make the principles and

applications of physical chemistry as clear as possible Proceeds from fundamental principles or postulates and shows how the consequences of these principles and postulates apply to the chemical and physical phenomena being studied Encourages the student not only to know the applications in physical chemistry but to understand where they come from Treats all topics relevant to undergraduate physical chemistry      **Western Medical Times** George Lee Servoss, 1926      **Journal of the American Chemical Society** American Chemical Society, 1928 Proceedings of the Society are included in v 1 59 1879 1937  
     **The Chemical News and Journal of Physical Science** , 1910      *Physical Chemistry in the Service of Medicine* Wolfgang Pauli, Martin Fischer, 1907      **Quantum Mechanics in Chemistry** Melvin W. Hanna, 1965      *Chemical News and Journal of Industrial Science* , 1910      **The Chemical News and Journal of Industrial Science** , 1913      *The Journal of Physical Chemistry* Wilder Dwight Bancroft, 1909 Includes section New Books      **The Journal of Physical Chemistry** , 1909      *Principles of Physical Chemistry* Hans Kuhn, Horst-Dieter Försterling, David H. Waldeck, 2009-03-17 Principles of Physical Chemistry Second Edition uniquely uses simple physical models as well as rigorous treatments for understanding molecular and supramolecular systems and processes In this way the presentation assists students in developing an intuitive understanding of the subjects as well as skill in quantitative manipulations The unifying nature of physical chemistry is emphasized in the book by its organization beginning with atoms and molecules and proceeding to molecular assemblies of increasing complexity ending with the emergence of matter that carries information i e the origin of life a physicochemical process of unique importance The aim is to show the broad scope and coherence of physical chemistry      *The Elements of Physical Chemistry* John Livingston Rutgers Morgan, 1908

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