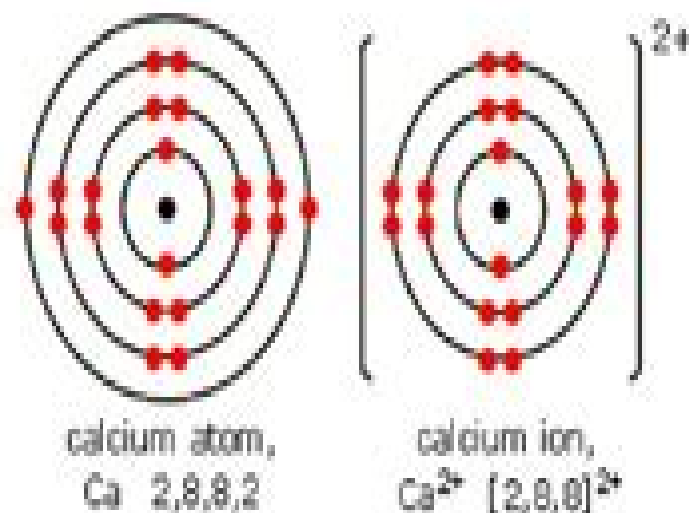
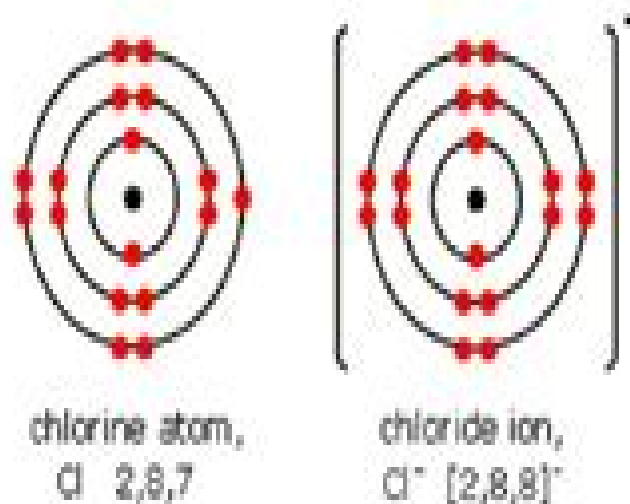


metal ions vs. non-metal ions

Metals will often **lose** their valence electrons to be left with a full valence shell. Atoms of metals **lose** e^- to form **positive** ions (cations).



Non-metals will **gain** electrons to fill their valence shell. Atoms of non-metals **gain** e^- to form **negative** ions (anions).



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