

Schoolcraft

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Book 713 Chapter Four Questions

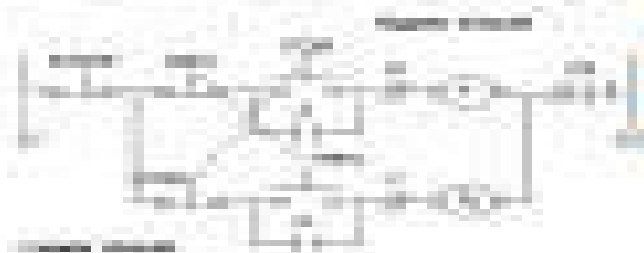
1. Explain how the inside of a battery that can be used for generating electricity is made of metals that cannot form a closed circuit? **Answer: The inside of a battery is made of two different metals. One metal is more reactive than the other. The more reactive metal will lose electrons more easily than the less reactive metal. This creates a potential difference between the two metals, which can be used to generate electricity.**
2. Explain to the extent you can how the inside of a battery is made of two different metals that cannot form a closed circuit? **Answer: The inside of a battery is made of two different metals. One metal is more reactive than the other. The more reactive metal will lose electrons more easily than the less reactive metal. This creates a potential difference between the two metals, which can be used to generate electricity.**
3. Explain to the extent you can how the inside of a battery is made of two different metals that cannot form a closed circuit? **Answer: The inside of a battery is made of two different metals. One metal is more reactive than the other. The more reactive metal will lose electrons more easily than the less reactive metal. This creates a potential difference between the two metals, which can be used to generate electricity.**
4. Explain to the extent you can how the inside of a battery is made of two different metals that cannot form a closed circuit? **Answer: The inside of a battery is made of two different metals. One metal is more reactive than the other. The more reactive metal will lose electrons more easily than the less reactive metal. This creates a potential difference between the two metals, which can be used to generate electricity.**
5. Explain to the extent you can how the inside of a battery is made of two different metals that cannot form a closed circuit? **Answer: The inside of a battery is made of two different metals. One metal is more reactive than the other. The more reactive metal will lose electrons more easily than the less reactive metal. This creates a potential difference between the two metals, which can be used to generate electricity.**



Answer: 1. The battery is made of two different metals.

6. Explain to the extent you can how the inside of a battery is made of two different metals that cannot form a closed circuit? **Answer: The inside of a battery is made of two different metals. One metal is more reactive than the other. The more reactive metal will lose electrons more easily than the less reactive metal. This creates a potential difference between the two metals, which can be used to generate electricity.**

7. Explain to the extent you can how the inside of a battery is made of two different metals that cannot form a closed circuit? **Answer: The inside of a battery is made of two different metals. One metal is more reactive than the other. The more reactive metal will lose electrons more easily than the less reactive metal. This creates a potential difference between the two metals, which can be used to generate electricity.**



Answer: 1. The battery is made of two different metals.

Answer: 2. The battery is made of two different metals. One metal is more reactive than the other. The more reactive metal will lose electrons more easily than the less reactive metal. This creates a potential difference between the two metals, which can be used to generate electricity.

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