

Introduction to Organic chemistry

Organic chemistry is the branch of chemistry that studies carbon compounds present in living things, once living things or synthetic/man-made.

Compounds that makes up living things whether alive or dead mainly contain carbon. Carbon is tetravalent.

It is able to form stable covalent bonds with itself and many non-metals like hydrogen, nitrogen ,oxygen and halogens to form a variety of compounds. This is because:

- (i) carbon uses all the four valence electrons to form four strong covalent bond.
- (ii)carbon can covalently bond to form a single, double or triple covalent bond with itself.
- (iii)carbon atoms can covalently bond to form a very long chain or ring.

When carbon covalently bond with Hydrogen, it forms a group of organic compounds called **Hydrocarbons**

A.HYDROCARBONS (HCs)

Hydrocarbons are a group of organic compounds containing /made up of hydrogen and carbon atoms only.

Depending on the type of bond that exist between the individual carbon atoms, hydrocarbon are classified as:

- (i) Alk**an**es
- (ii) Alk**en**es
- (iii) Alk**yn**es

(i) Alk**an**es

(a)Nomenclature/Naming

These are hydrocarbons with a general formula **C_nH_{2n+2}** where **n** is the number of **Carbon** atoms in a molecule.

The carbon atoms are linked by single bond to each other and to hydrogen atoms.

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