

- Organic reductions often involve hydrogen ions
- $R + nH^+ + ne \xrightarrow{\text{clock}} RH_n$
- Good buffering reqd in organic polarography

Metal Complexes

- $ML_p + ne + Hg \xrightarrow{\text{clock}} M(Hg) + pL$
- Difference between half-wave potential for complexed and uncomplexed metal ion is given by:
- $E_{1/2(c)} - E_{1/2(\text{free})} =$

$$RT/nF \ln K_d - RT/nF p \ln [L] + RT/nF \ln (D_{\text{free}}/D_{(c)})^{1/2}$$

- Stoichiometry can thus be determined by plotting $E_{1/2}$ versus $[L]$
- Also possible to improve resolution between neighbouring waves by carefully choosing ligand and concentration

Polarography Of Metal Complexes

SA Dillow



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