Molecular Geometry

	T-1				
Steric Number	Lone Pairs = 0	Lone Pairs = 1	Lone Pairs = 2	Lone Pairs = 3	Lone Pairs = 4
2	Linear				
3	Trigonal Planar	Angular or Bent			
4:	109.5* Tetrahedral	Trigonal Pyramidal	Angular or Bent		
5	120° 90° 120° Trigonal Bipyramidal	Seesaw	T-shaped	Linear	
6	90° 180° Octahedral	Square Pyramidal	Square Planar	T-shaped	Linear

David Kirk

Unveiling the Magic of Words: A Overview of "Shape In Chemistry An Introduction To Molecular Shape And Topology"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Shape In Chemistry An Introduction To Molecular Shape And Topology**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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