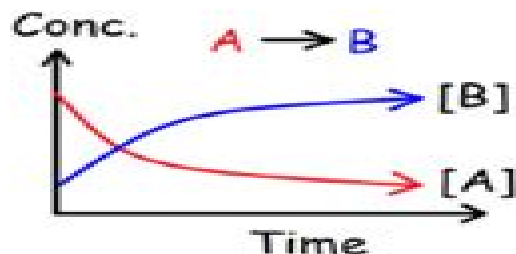


## Chemical Kinetics:



**Instantaneous Rate of Appearance:**

$$\text{Rate} = + \frac{d[B]}{dt}$$

**Method of Initial Rates:**

Trial	[A]	[B]	[C]	I. Rate
1	0.10 M	0.10 M	0.10 M	0.20 M/s
2	0.20 M	0.10 M	0.10 M	0.40 M/s
3	0.10 M	0.20 M	0.10 M	0.80 M/s
4	0.10 M	0.10 M	0.20 M	0.20 M/s

**Rate Constant k:**

$$k = \frac{\text{Rate}}{[A]^x[B]^y[C]^z}$$

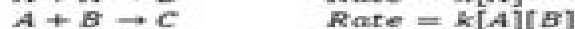
**Units of k:**  $M^{1-n} t^{-1}$  or  $(mol)^{1-n} (L)^{n-1} t^{-1}$

**Note:**  $M = mol \cdot L^{-1}$  and  $t \rightarrow s, min, hr, days$

**Unimolecular:**



**Bimolecular:**



**Termolecular:**



**Average Rate of Appearance:**  $A \rightarrow B$

$$\text{Rate} = + \frac{\Delta[B]}{\Delta t} = \frac{[B]_f - [B]_i}{t_f - t_i}$$

**Average Rate of Disappearance:**

$$\text{Rate} = - \frac{\Delta[A]}{\Delta t}$$

**Rate of a Chemical Reaction:**



$$\text{Rate} = - \frac{1}{2} \frac{\Delta[A]}{\Delta t} = - \frac{1}{3} \frac{\Delta[B]}{\Delta t} = + \frac{1}{4} \frac{\Delta[C]}{\Delta t} = + \frac{1}{5} \frac{\Delta[D]}{\Delta t}$$

**Differential Rate Law Expression:**

$$\text{Rate} = k[A]^x[B]^y[C]^z$$

**Finding The Order of a Reactant:**

$$x = \frac{\log \left( \frac{\text{Rate 2}}{\text{Rate 1}} \right)}{\log \left( \frac{[A]_2}{[A]_1} \right)} \quad y = \frac{\log \left( \frac{\text{Rate 3}}{\text{Rate 1}} \right)}{\log \left( \frac{[B]_3}{[B]_1} \right)}$$

**Overall order of the reaction:**

$$\text{Order} = x + y + z$$

**Factors Affecting the Rate of a Reaction:**

1. Temperature
2. Concentration
3. Catalyst
4. Surface Area
5. The Nature of the Reactants

**Reaction Mechanisms:**



**Note:**  $\text{Catalyst} \rightarrow B$      $\text{Intermediate} \rightarrow D$

# Research In Chemical Kinetics V 1 Research In Chemical Kinetics

**Jicheng Xie**



**Research In Chemical Kinetics V 1 Research In Chemical Kinetics:**

## Decoding **Research In Chemical Kinetics V 1 Research In Chemical Kinetics**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Research In Chemical Kinetics V 1 Research In Chemical Kinetics**," a mesmerizing literary creation penned by a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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