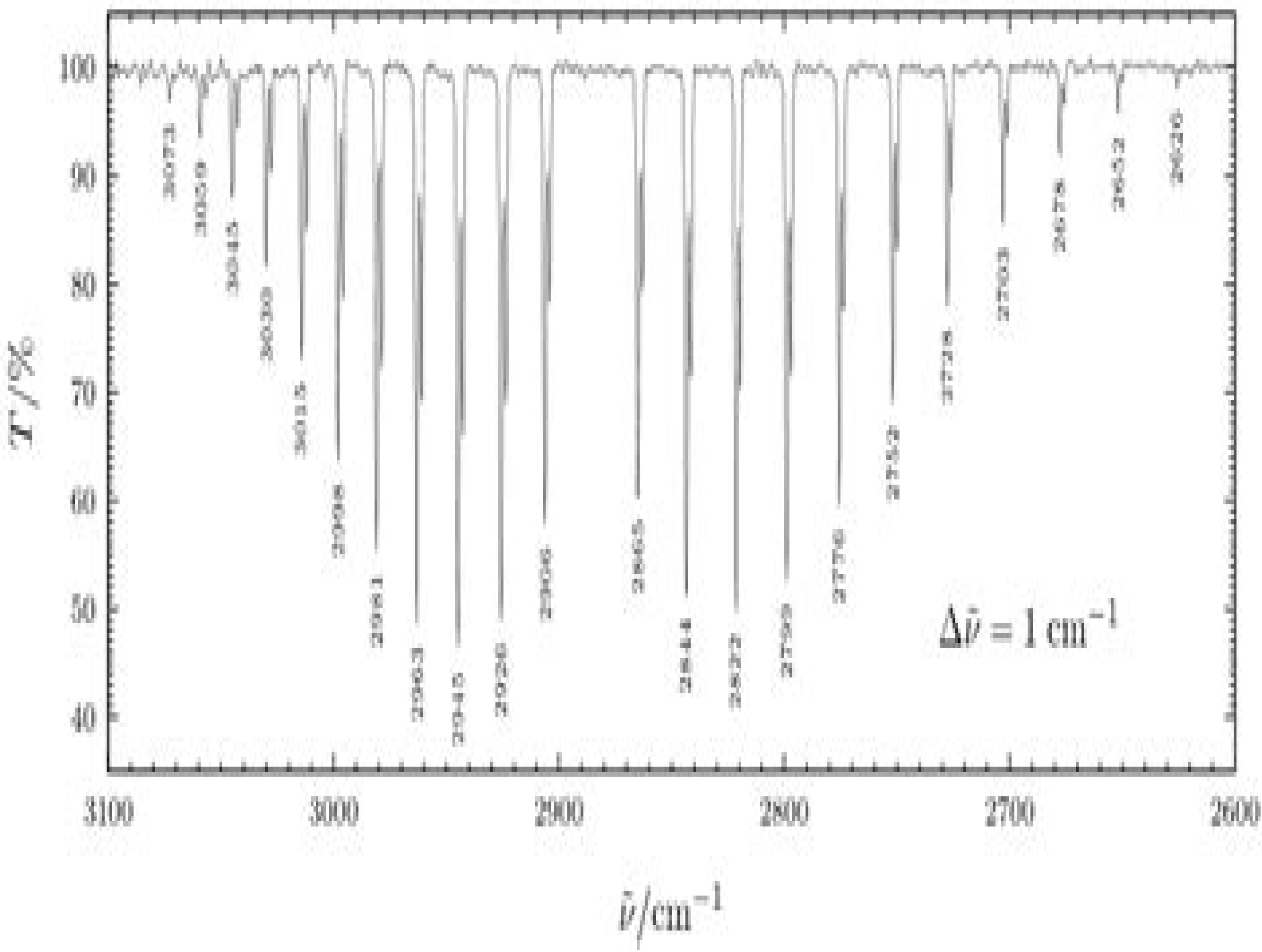




Rotational Spectroscopy Of Diatomic Molecules

John M. Brown, Alan Carrington



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discreet bundles to the resultant black body radiation curve a body of knowledge has developed with profound consequences in our ability to understand nature In the early years quantum theory was the providence of physicists and certain breeds of physical chemists While physicists honed and refined the theory and studied atoms and their component systems physical chemists began the foray into the study of larger molecular systems Quantum theory predictions of these systems were first verified through experimental spectroscopic studies in the electromagnetic spectrum microwave infrared and ultraviolet visible and later by nuclear magnetic resonance NMR spectroscopy Over two generations these studies were hampered by two major drawbacks lack of resolution of spectroscopic data and the complexity of calculations This powerful theory that promised understanding of the fundamental nature of molecules faced formidable challenges The following example may put things in perspective for today s chemistry faculty college seniors or graduate students As little as 40 years ago force field calculations on a molecule as simple as ketene was a four to five year dissertation project Basic Atomic and Molecular Spectroscopy John Michael Hollas,2002 The latest in the Tutorial Chemistry Texts series Basic Atomic and Molecular Spectroscopy contains chapters on quantization in polyelectronic atoms molecular vibrations and electronic spectroscopy

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