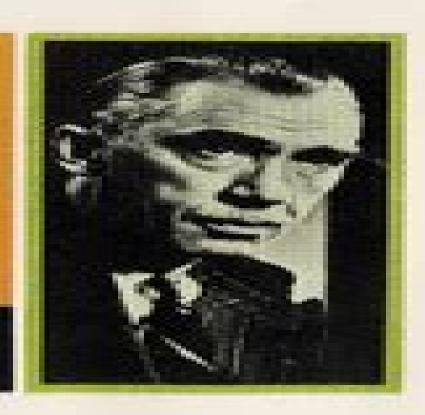
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web 1 draw five protons in the nucleus of the atom label them with their charge 2 draw six neutrons in the nucleus of the atom 3 draw two electrons in the first energy level and label them with their charge 4 draw three electrons in the second energy level and label them with their charge 5 what element is represented by the diagram

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web may 18 2021 figure pageindex 1 the structure of the atom atoms have protons and neutrons in the center making the nucleus while the electrons orbit the nucleus the modern atomic theory states that atoms of one element are the same while atoms of different elements are different

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web a carbon 12 12 c b this atom contains six protons and six neutrons there are six electrons in a neutral 12 c atom the net charge of such a neutral atom is zero and the mass number is 12 c the preceding answers are correct d the atom will be stable since c 12 is a stable isotope of carbon e the preceding answer is correct

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web the atomic number tells you the number of of an n one atom of an element it also tells you the number of in a neutral atom of that element the atomic number gives the identity of an element as well as ts location on the periodic table no two different elements will have the give the symbol and number of protons in one atom of atomic number

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neutrons to get an answer of 9 2 the lowest principal quantum number for an electron is 0 1 2 3

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web a both the electron geometry and the molecular structure are octahedral b both the electron geometry and the molecular structure are trigonal bipyramid c both the electron geometry and the molecular structure are linear d both the electron geometry and the molecular structure are trigonal planar

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web exams prep master atomic structure incorporates three elements that wholly make an atom and they are named electron proton and neutron in the structure of an atom the nucleus sets in the center and it contains the proton whereas electrons lie on the periphery of the atom in an orbital way the first atomic model was given by j j thomson

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