

1. Consider the following reaction:



An equilibrium mixture of this reaction at a certain temp was found to have $[\text{CO}] = 0.115\text{M}$, $[\text{H}_2] = 0.110\text{M}$, and $[\text{CH}_3\text{OH}] = 0.190\text{M}$. What is the value of the equilibrium constant (K_c) at this temperature?

$$K_c = \frac{[\text{CH}_3\text{OH}]}{[\text{CO}][\text{H}_2]^2} = \frac{(0.19)}{(0.115)(0.11)^2} = 136.5$$

2. Consider the following reaction:



An equilibrium mixture of this reaction at a certain temperature was found to have $[\text{NH}_3] = 0.274\text{M}$ and $[\text{H}_2\text{S}] = 0.355\text{M}$. What is the value of the equilibrium constant (K_c) at this temperature?

$$K_c = [\text{NH}_3][\text{H}_2\text{S}] = (0.274)(0.355) = 0.0973$$

3. Consider the following reaction:



The reaction between CO and H_2 is carried out at a specific temperature with initial concentrations of $\text{CO} = 0.32\text{M}$ and $\text{H}_2 = 0.45\text{M}$. At equilibrium, the concentration of CH_3OH is 0.14M . Find the equilibrium constant at this temperature.

	CO	2H_2	CH_3OH
I	0.32	0.45	
C	-x	-2x	+x
E	0.18	0.17	0.14

$$K_c = \frac{(0.14)}{(0.18)(0.17)^2} = 26.9$$

4. Consider the following reaction:



The reaction of CH_4 is carried out at some temperature with an initial concentration of $[\text{CH}_4] = 0.090\text{M}$. At equilibrium, the concentration of H_2 is 0.012M . Find the equilibrium constant at this temperature.

	2CH_4	C_2H_2	3H_2
I	0.09		
C	-2x	+x	+3x
E	0.082	0.004	0.012M

$$K_c = \frac{(0.004)(0.012)^3}{(0.082)^2} = 1.0 \times 10^{-6}$$

5. Consider the following reaction:



A reaction mixture contains $[\text{NO}_2] = 0.0248\text{M}$ and $[\text{N}_2\text{O}_4] = 0.0337\text{M}$. Calculate Q_c at the given concentrations.

$$Q_c = \frac{(0.0248)^2}{(0.0337)} = 0.0183 \quad Q_c > K_c, \text{ rxn goes left}$$

6. The following reaction was carried out in a 2.75L reaction vessel at 1100K :



Mastering Chemistry Problems

Gerald W. Gibson



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Mastering Chemistry Problems Gerald W. Gibson, Gary C. Faber, 1988-01-01 Mastering Chemistry Gerald W. Gibson, 1975 **Chemistry** John Olmsted, Gregory M. Williams, 1997 Textbook outlining concepts of molecular science

EBOOK: GENERAL CHEMISTRY, THE ESSENTIAL CONCEPTS CHANG, 2013-01-07 EBOOK GENERAL CHEMISTRY THE ESSENTIAL CONCEPTS *Teaching in the Sciences* Acram Taji, 2005-02-14 Gain a clear understanding of what effective teachers do and how successful students learn Over the past 20 years a greater concentration on research aimed at both teaching and learning has revealed that chalk and talk teaching copying notes and cookbook practical lessons offer little challenge to student Problems and Problem Solving in Chemistry Education Georgios Tsapalis, 2021-05-17 Problem solving is central to the teaching and learning of chemistry at secondary tertiary and post tertiary levels of education opening to students and professional chemists alike a whole new world for analysing data looking for patterns and making deductions As an important higher order thinking skill problem solving also constitutes a major research field in science education Relevant education research is an ongoing process with recent developments occurring not only in the area of quantitative computational problems but also in qualitative problem solving The following situations are considered some general others with a focus on specific areas of chemistry quantitative problems qualitative reasoning metacognition and resource activation deconstructing the problem solving process an overview of the working memory hypothesis reasoning with the electron pushing formalism scaffolding organic synthesis skills spectroscopy for structural characterization in organic chemistry enzyme kinetics problem solving in the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner *Chemistry* Theodore E. Brown, Theodore Lawrence Brown, H. Eugene H. LeMay, Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Matthew E. Stoltzfus, 2014 ALERT Before you purchase check with your instructor or review your course syllabus to ensure that you select the correct ISBN Several versions of Pearson's MyLab check with the seller before completing your purchase Used or rental books If you rent or purchase a used book with an access code the access code may have been redeemed previously and you may have to purchase a new access code Access codes Access codes that are purchased from sellers other than Pearson carry a higher risk of being either the wrong ISBN or a previously redeemed code Check with the seller prior to purchase The trusted innovative calibrated leader now tightly integrated with MasteringChemistry Unrivaled problems notable scientific accuracy and currency and remarkable clarity have made Chemistry The Central Science the leading general chemistry text for more than a decade Trusted innovative and calibrated the text increases conceptual understanding and student success in general chemistry by building on the expertise of the dynamic author team of leading researchers and award winning professors The

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Ways to Score Higher on Your MCAT Marti Anne Maguire,Paula Stiles,2010 Contains over one hundred tips and suggestions to improve SAT scores with step by step instructions to write better essays and reviews of grammar usage and math concepts details a test taking strategy that encourages time management and educated guessing and provides resources for practice tests **Selected Papers of the Association for Teacher Education in Europe Spring**

Conference 2015 Linda Daniela,Irēna Žogla,2016-12-14 The Association for Teacher Education in Europe ATEE is a non profit European organisation with the aim of enhancing the quality of teacher education in Europe and supporting the professional development of teachers and teacher educators at all levels The contents of this book represent the best papers presented at the 2015 ATEE Spring Conference a forum at which European teacher educators could learn from each other and also from experiences research and ideas from outside Europe As such this book will serve to stimulate dialogue among

teacher educators researchers on teacher education students teachers employers politicians supervisory bodies NGOs and other groups involved in teacher education and research and innovation in teacher education Chemistry Education and Sustainability in the Global Age Mei-Hung Chiu,Hsiao-Lin Tuan,Hsin-Kai Wu,Jing-Wen Lin,Chin-Cheng Chou,2012-12-05 This edited volume of papers from the twenty first International Conference on Chemical Education attests to our rapidly changing understanding of the chemistry itself as well as to the potentially enormous material changes in how it might be taught in the future Covering the full range of appropriate topics the book features work exploring themes as various as e learning and innovations in instruction and micro scale lab chemistry In sum the 29 articles published in these pages focus the reader s attention on ways to raise the quality of chemistry teaching and learning promoting the public understanding of chemistry deploying innovative technology in pedagogy practice and research and the value of chemistry as a tool for highlighting sustainability issues in the global community Thus the ambitious dual aim achieved in these pages is on the one hand to foster improvements in the teaching and communication of chemistry whether to students or the public and secondly to promote advances in our broader understanding of the subject that will have positive knock on effects on the world s citizens and environment In doing so the book addresses as did the conference the neglect suffered in the chemistry classroom by issues connected to globalization even as it outlines ways to bring the subject alive in the classroom through the use of innovative technologies **Organic Chemistry** Allan D. Headley,2020-01-02 Provides an in depth study of organic compounds that bridges the gap between general and organic chemistry Organic Chemistry Concepts and Applications presents a comprehensive review of organic compounds that is appropriate for a two semester sophomore organic chemistry course The text covers the fundamental concepts needed to understand organic chemistry and clearly shows how to apply the concepts of organic chemistry to problem solving In addition the book highlights the relevance of organic chemistry to the environment industry and biological and medical sciences The author includes multiple choice questions similar to aptitude exams for professional schools including the Medical College Admissions Test MCAT and Dental Aptitude Test DAT to help in the preparation for these important exams Rather than categorize content information by functional groups which often stresses memorization this textbook instead divides the information into reaction types This approach bridges the gap between general and organic chemistry and helps students develop a better understanding of the material A manual of possible solutions for chapter problems for instructors and students is available in the supplementary websites This important book Provides an in depth study of organic compounds with division by reaction types that bridges the gap between general and organic chemistry Covers the concepts needed to understand organic chemistry and teaches how to apply them for problem solving Puts a focus on the relevance of organic chemistry to the environment industry and biological and medical sciences Includes multiple choice questions similar to aptitude exams for professional schools Written for students of organic chemistry Organic Chemistry Concepts and Applications is the comprehensive text that presents the

material in clear terms and shows how to apply the concepts to problem solving *MasteringChemistry with Pearson EText -- Standalone Access Code Card -- for General Chemistry* John C. McMurry, Robert C. Fay, John E. McMurry, 2013-01-03

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such as training regimes and formal education access to tools socialisation and cultural understanding budgetary constraints and market demands changing technologies and materials and political and economic regimes this book sheds fresh light on the multifarious forms of intelligence involved in design and making inventing and manufacturing and cultivating and producing As such it will appeal to scholars of anthropology sociology and cultural geography as well as to craftspeople with interests in creativity skilful practice perception and ethnography **Mastering Organic Chemistry** Gerald W.

Gibson,1979 General Chemistry Ralph H. Petrucci,William S. Harwood,1993 General Chemistry Principles and Modern Applications is recognized for its superior problems lucid writing and precision of argument This updated and expanded edition retains the popular and innovative features of previous editions including Feature Problems follow up Integrative and Practice Exercises to accompany every in chapter Example and Focus On application boxes as well as new Keep in Mind marginal notes Topics covered include atoms and the atomic theory chemical compounds and reactions gases Thermochemistry electrons in atoms chemical bonding liquids solids and intermolecular forces chemical kinetics principles of chemical equilibrium acids and bases electrochemistry representative and transitional elements and nuclear and organic chemistry For individuals interested in a broad overview of chemical principles and applications **Broadening**

Participation in STEM Zayika Wilson-Kennedy,Goldie S. Byrd,Eugene Kennedy,Henry T. Frierson,2019-02-28 This book reports on high impact educational practices and programs that have been demonstrated to be effective at broadening the participation of underrepresented groups in the STEM disciplines **Problem-Based Learning for Math & Science**

Diane L. Ronis,2008 Teachers looking for a concise guide to implementing problem based learning in math and science classrooms This book is for you Debra Gerdes Professional Development Leader Illinois Mathematics and Science Academy The purpose of problem based learning is to emphasize meaning making over fact collecting With this method Diane Ronis has written a book that is well equipped to produce self motivated and independent lifelong learners Katie Morrow Technology Integration Specialist O Neill Public Schools NE Increase students skills and content retention in math and science What s the best way to create a real world instructional environment where students are involved in firsthand experiences and where important ideas are connected to meaningful life events that help deepen learners understanding Diane Ronis demonstrates how the problem based learning PBL method gives students the opportunity to actively explore and resolve authentic problem simulations and student identified problems in the community while strengthening their problem solving skills Updated throughout this second edition illustrates how to use the PBL inquiry process with Internet resources to create an integrated instructional environment and also provides Problem based learning activities relating to math and science in each chapter Projects that correlate to national science mathematics and technology standards Student handouts evaluation forms and all the information necessary for successful project completion Problem Based Learning for Math and Science Second Edition is the perfect resource for educators who want to expand their teaching repertoire and

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