Canada balsam is then added, first round the blood and then to fill up the space. The layer of blood must not be too thick. Canada balsam which is yellow and not quite clear is best suited for the purpose. The blood must remain for a few days exposed to the air, that is until crystallisation has ceased and the odour of the balsam vanished. The excess of balsam is then removed with the help of a knife, wetted with ether, turpentine, or oil of cloves; the preparation is then covered with a glass, which is fixed on with asphalt or balsam.

A. K. M.

Study of Metahæmoglobin. By A. Jäderholm (Zeit. f. Biol., 20, 419—448).—For the purpose of obtaining metahæmoglobin crystals, the author treated dog's blood, with slight modifications, according to the sixth process described in Prever's "Die Blutkrystalle." The crystals differ only in size and shape from those Hammarsten obtained by treating horse's blood with ferricyanide of potassium and subsequent dilution and application of cold (Zeit. Physiol. Chem., 8, 186), as both forms of crystals and their solutions give the characteristic spectra of metahæmoglobin.

The addition of a very small quantity of sodium carbonate (0.00053 per cent.) to a solution, produces the so-called alkaline metahemoglobin spectrum $\pi + \alpha_1 + \beta_1$, π being always weaker than $\alpha_1 + \beta_1$.

The bands II and III of metahæmoglobin, and $\alpha_1 + \beta_1$, of its alkaline solution, correspond very nearly with $\alpha + \beta$ of oxyhæmoglobin. The spectrum $\pi + \alpha_1 + \beta_1$ of alkaline metahæmoglobin can be obtained also by passing pure hydrogen through a solution.

If excess of sodium carbonate be avoided on the one hand, and of hydrogen on the other, the bands I and IV of metahæmoglobin are produced on shaking the solutions with air.

The latter half of the paper is devoted to the discussion of its constitution. The author explains his reason for assuming it to be a peroxidised oxyhemoglobin, as he did in his former paper, but now agrees with Hüfner and Hulz in considering it to contain the same amount of oxygen as oxyhemoglobin.

J. P. L.

Physiological Chemistry.

Influence of Variations in the Percentage Composition of the Air on the Intensity of Respiratory Changes. By L. Frederic (Compt. rend., 99, 1124—1125).—The author has investigated the influence of variations in the amount of carbonic anhydride or of oxygen in the air on the intensity of respiratory changes, as measured by the amount of oxygen absorbed in the case of rabbits and of himself.

An increase in the proportion of oxygen has no effect on the absorption of oxygen in the process of respiration. When an animal is

Review Of Physiological Chemistry

Mark Steyvers

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Table of Contents Review Of Physiological Chemistry

- 1. Understanding the eBook Review Of Physiological Chemistry
 - The Rise of Digital Reading Review Of Physiological Chemistry
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Review Of Physiological Chemistry
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Review Of Physiological Chemistry
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Review Of Physiological Chemistry
 - Personalized Recommendations
 - Review Of Physiological Chemistry User Reviews and Ratings
 - Review Of Physiological Chemistry and Bestseller Lists

- 5. Accessing Review Of Physiological Chemistry Free and Paid eBooks
 - Review Of Physiological Chemistry Public Domain eBooks
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 - Review Of Physiological Chemistry Budget-Friendly Options
- 6. Navigating Review Of Physiological Chemistry eBook Formats
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 - Review Of Physiological Chemistry Compatibility with Devices
 - Review Of Physiological Chemistry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Review Of Physiological Chemistry
 - Highlighting and Note-Taking Review Of Physiological Chemistry
 - Interactive Elements Review Of Physiological Chemistry
- 8. Staying Engaged with Review Of Physiological Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Review Of Physiological Chemistry
- 9. Balancing eBooks and Physical Books Review Of Physiological Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Review Of Physiological Chemistry
- 10. Overcoming Reading Challenges
 - o Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Review Of Physiological Chemistry
 - Setting Reading Goals Review Of Physiological Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Review Of Physiological Chemistry
 - Fact-Checking eBook Content of Review Of Physiological Chemistry
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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