



Louis Riel

**G Psacharopoulos**



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 $\text{NO}_3^- \text{aq}$   $\text{K}^+ \text{aq}$   $\text{NO}_3^- \text{aq}$   $\text{H}_2\text{O}$  l answer  $\text{K}^+ \text{aq}$  and  $\text{NO}_3^- \text{aq}$  identify the chemical equation the spectator ions  $\text{Pb}^{2+} \text{aq}$   $2 \text{KI} \text{aq}$   $\text{PbI}_2 \text{s}$   
 $2 \text{KNO}_3 \text{aq}$

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web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions  
gives the net ionic equation which shows only those species that participate in the chemical reaction  $2 \text{Ag}^+ \text{aq}$   $\text{CrO}_4^{2-} \text{aq}$   
 $\rightarrow \text{Ag}_2\text{CrO}_4 \text{s}$  label 4 2 3

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and sulfate ion are the spectator ions in this reaction they appear unchanged in both the product and reactant side of the  
equation

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web mar 11 2014 solution ionic equation  $\text{H}^+ \text{aq}$   $\text{Cl}^- \text{aq}$   $\text{Na}^+ \text{aq}$   $\text{HCO}_3^- \text{aq}$   $\text{Na}^+ \text{aq}$   $\text{Cl}^- \text{aq}$   $\text{H}_2\text{O}$   $\text{l}$   $\text{CO}_2 \text{g}$  the spectator ions are  $\text{Cl}^-$  and  $\text{Na}^+$

net ionic equation  $\text{H}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{Na}^+ \text{aq} + \text{HCO}_3^- \text{aq} + \text{Na}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g}$  here is a video on spectator ions  
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web the six  $\text{NO}_3^- \text{aq}$  ions and the six  $\text{Na}^+ \text{aq}$  ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation  $3\text{Ba}^{2+} \text{aq} + 2\text{PO}_4^{3-} \text{aq} \rightarrow \text{Ba}_3(\text{PO}_4)_2 \text{s}$

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water  $\text{H}^+ \text{aq} + \text{OH}^- \text{aq} \rightarrow \text{H}_2\text{O} \text{l}$  the spectator ions in  $\text{KOH} \text{aq}$  and  $\text{HNO}_3 \text{aq}$  are  $\text{K}^+ \text{aq}$  and  $\text{NO}_3^- \text{aq}$  how to identify and list the

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web aug 8 2022 a spectator ion is an ion that does not take part in the chemical reaction and is found in solution both before and after the reaction the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction

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