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1994-1995



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 $3\text{Ca(OH)}_2 + 1\text{Al}_2\text{SO}_4 \rightarrow 3\text{CaSO}_4 + 2\text{Al(OH)}_3$ type of reaction double displacement
 $3\text{Mg} + 1\text{Fe}_2\text{O}_3 \rightarrow 2\text{Fe} + 3\text{MgO}$ type of reaction single displacement
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 $2\text{Fe} + 3\text{MgO} \rightarrow 2\text{FeO} + 3\text{Mg}$ type of reaction double displacement
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 $\text{Ca(OH)}_2 + \text{MgSO}_4 \rightarrow \text{CaSO}_4 + \text{Mg(OH)}_2$ type of reaction double displacement
 $\text{C}_2\text{H}_4 + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O}$ type of reaction single displacement
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