

The body performs many processes that depend upon energy and could not occur without a supply of ATP. How does the energy from ATP power the cellular reactions?

- > a. The cellular processes are coupled to hydrolysis of ATP, an exergonic process
- b. Cellular processes are coupled to hydrolysis of ATP, an endergonic process
- c. Energy from ATP hydrolysis is stored in cellular chambers
- d. ATP hydrolysis absorbs the remaining energy from formation of glucose-6-phosphate
- e. All of the energy from hydrolysis of ATP is released as heat

Using the following table of free energies:

ATP \rightarrow ADP + Pi $\Delta G = -7.3$ kcal/mol

Glucose-1-phosphate \rightarrow Glucose + Pi $\Delta G = -5.0$ kcal/mol

Find the free energy change for the reaction: Glucose + ATP \rightleftharpoons glucose-1-phosphate + ADP

- a. $\Delta G = -23.0$ kcal/mol
- b. $\Delta G = +23.0$ kcal/mol
- > c. $\Delta G = -2.3$ kcal/mol
- d. $\Delta G = +2.3$ kcal/mol
- e. $\Delta G = 0$

A human being needs a mass of ATP nearly equivalent to his or her own body mass daily. What process allows a ready supply of ATP for cellular reactions?

- a. The body stores large amounts of ATP in mitochondria
- b. The body stores large amounts of ATP in adipocytes
- c. Cellular nuclei can make ATP rapidly from air
- > d. Cellular processes make new ATP as fast as it is used
- e. ATP is never actually used but must only be present for reactions to occur

Ribose can best be described as an:

- a. Kethexose
- > b. Aldopentose
- c. Ketopentose
- d. Ketotriose
- e. Kethexptose

The carbohydrate known as milk sugar is also known by the common name:

- a. Sucrose
- b. Fructose
- > c. Lactose
- d. Galactose
- e. Maltose

Shr Princ Biochem Txt Lecture Outlines

**Eric E. Conn, Eric Edward Conn, Paul K.
Stumpf**



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